

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme betreffen die Veranschaulichung des Verhaltens der Stoffeigenschaften mittels der Parameter der Zustandsgleichungen. In der nachfolgenden Tabelle sind die Parameter der Zustandsgleichungen angegeben. Die Stoffeigenschaften sind in der Tabelle angegeben.

$$p = p(T, V) = p(T, \rho)$$

1. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

$$T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K}$$

2. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

$$T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K}$$

3. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

$$T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K}$$

4. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

$$T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K}$$

5. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

$$T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K} \quad \text{oder} \quad T = 300 \text{ K}$$

6. Ein Stoff wird in einem Zylinder bei einer Temperatur von 300 K und einem Volumen von 1 m³ auf einen Druck von 10 bar erhitzt. Wie hoch ist die Temperatur des Stoffes?

Section 3 Gases Ch 11 Review Answers

Jin-Ying Zhang



Section 3 Gases Ch 11 Review Answers:

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