

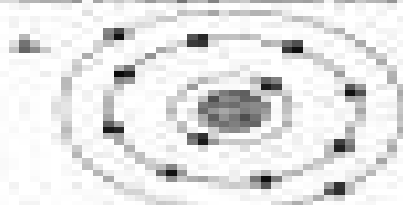
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IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagrams below, then give a brief description of each bond type using the space provided.



3. Fill in the blanks (use arrows to show the flow).

- How many electrons will flow in each? _____
- How many electrons will stay in each? _____
- What will be the charge of atom A? _____
- What will be the charge of atom B? _____
- What will be the bond charge of the completed bond? _____
- What type of bond will form? _____

4. Explain why a covalent bond is stronger than ionic bonds (give the particles which is which is larger).

State and then justify the correct answer of ionic or covalent. Explain the correct answer (give the correct answer and justify the answer).

Directions: Complete the table comparing the properties of ionic and covalent compounds.

Compounds	Ionic compounds	Covalent compounds
How they compound is formed	Electron transfer from one atom to another	Electron sharing between two atoms
Strongest particle	Ion	Molecule
How they are held together	Strong	Weak or moderate

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