



Lab Notebook

Classifying Chemical Reactions

Directions: Write the type of chemical reaction in the space provided next to each equation. Name the product or the products of the reaction at the end.

Column I

1. _____ 1. A precipitate forms in a gas liquid collection over water (combustion reactions are prohibited)
2. _____ 2. Two reactants exchange products and/or anions/cations
3. _____ 3. A hydrocarbon explodes, usually in a combustion
4. _____ 4. Chlorine atoms break down into many pieces (decomposition)
5. _____ 5. A type of synthesis reaction that produces one molecule

Directions: Write the correct type of chemical reaction in the space provided.

combustion _____ 6. $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

decomposition _____ 7. $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$

synthesis

single displacement

_____ 8. $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$

double displacement reaction _____ 9. $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$

synthesis reaction

_____ 10. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

_____ 11. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

decomposition

single displacement reaction

_____ 12. $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

synthesis reaction

double displacement reaction

_____ 13. $\text{H}_2 + \text{Cl}_2 \rightarrow \text{HCl}$

decomposition

double displacement reaction

_____ 14. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

double displacement reaction _____ 15. $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$

double displacement reaction

Section 2 Reinforcement Classifying Chemical Reactions

Patrick Vollmar



Section 2 Reinforcement Classifying Chemical Reactions:

Uncover the mysteries within Crafted by is enigmatic creation, Discover the Intrigue in **Section 2 Reinforcement Classifying Chemical Reactions** . This downloadable ebook, shrouded in suspense, is available in a PDF format (PDF Size: *). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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