

Chapter 2, The Chemistry of Life (continued)**Section 2–3 Carbon Compounds** (pages 44–48)

This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things.

The Chemistry of Carbon (page 44)

1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons.
2. What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms.

Macromolecules (page 45)

3. Many of the molecules in living cells are so large that they are known as macromolecules.
4. What is the process called by which macromolecules are formed? Polymerization
5. When monomers join together, what do they form? Polymers
6. What are four groups of organic compounds found in living things?
 - a. Carbohydrates
 - b. Lipids
 - c. Nucleic acids
 - d. Proteins

Carbohydrates (pages 45–46)

7. What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbohydrates.
8. Circle the letter of each sentence that is true about carbohydrates.
 - ☒ a. Starches and sugars are examples of carbohydrates.
 - ☒ b. Living things use them as their main source of energy.
 - ☐ c. The monomers in sugar polymers are starch molecules.
 - ☒ d. Plants and some animals use them for strength and rigidity.
9. Single sugar molecules are also called monosaccharides.
10. Circle the letter of each monosaccharide.
☒ a. galactose b. glycogen ☒ c. glucose ☒ d. fructose
11. What are polysaccharides? They are large macromolecules formed from monosaccharides.
12. How do plants and animals store excess sugar? Plants use a polysaccharide called plant starch, whereas animals use a polysaccharide called glycogen.

Section 2 3 Carbon Compounds Pages 44 48 Answers

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Table of Contents Section 2 3 Carbon Compounds Pages 44 48 Answers

1. Understanding the eBook Section 2 3 Carbon Compounds Pages 44 48 Answers
 - The Rise of Digital Reading Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 2 3 Carbon Compounds Pages 44 48 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Personalized Recommendations
 - Section 2 3 Carbon Compounds Pages 44 48 Answers User Reviews and Ratings
 - Section 2 3 Carbon Compounds Pages 44 48 Answers and Bestseller Lists

5. Accessing Section 2 3 Carbon Compounds Pages 44 48 Answers Free and Paid eBooks
 - Section 2 3 Carbon Compounds Pages 44 48 Answers Public Domain eBooks
 - Section 2 3 Carbon Compounds Pages 44 48 Answers eBook Subscription Services
 - Section 2 3 Carbon Compounds Pages 44 48 Answers Budget-Friendly Options
6. Navigating Section 2 3 Carbon Compounds Pages 44 48 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 2 3 Carbon Compounds Pages 44 48 Answers Compatibility with Devices
 - Section 2 3 Carbon Compounds Pages 44 48 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Highlighting and Note-Taking Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Interactive Elements Section 2 3 Carbon Compounds Pages 44 48 Answers
8. Staying Engaged with Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 2 3 Carbon Compounds Pages 44 48 Answers
9. Balancing eBooks and Physical Books Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 2 3 Carbon Compounds Pages 44 48 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Setting Reading Goals Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Fact-Checking eBook Content of Section 2 3 Carbon Compounds Pages 44 48 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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