



Chemical Changes

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula of hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula of water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula of oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in the medical field as a disinfectant and antiseptic. The reaction of hydrogen peroxide to form water and oxygen is a common example of a chemical change.

4. How many moles of hydrogen peroxide are there in 100 grams of hydrogen peroxide? How many moles of water and oxygen are there in 100 grams of hydrogen peroxide?

100 grams



5. What is the chemical formula of water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

6. What is the chemical formula of oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

7. What is the chemical formula of hydrogen peroxide? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in the medical field as a disinfectant and antiseptic.

8. What is the chemical formula of water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

9. What is the chemical formula of oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

In nature, the reaction of H_2O_2 to form H_2O and O_2 is a common example of a chemical change.

10. How many moles of hydrogen peroxide are there in 100 grams of hydrogen peroxide? How many moles of water and oxygen are there in 100 grams of hydrogen peroxide?

Section 1 Reinforcement Chemical Changes Chapter 24

Yijin Wang



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