

- 8 The gas in a can of hairspray is made up mainly of a hydrocarbon X. The boiling point of the hydrocarbon X is  $-5^{\circ}\text{C}$  and its melting point is  $-120^{\circ}\text{C}$ .

(a) Complete the table below by making use of the above information.

| temperature ( $^{\circ}\text{C}$ ) | state of hydrocarbon X |
|------------------------------------|------------------------|
| 30                                 |                        |
| -25                                | liquid                 |
| -60                                |                        |

[1]

- (b) If 'O' represents a molecule of hydrocarbon X, show the arrangement of the molecules at  $-25^{\circ}\text{C}$  and at  $30^{\circ}\text{C}$ , in the boxes below.



$-25^{\circ}\text{C}$



$30^{\circ}\text{C}$

[2]

- (c) Hydrocarbon X is usually kept as a liquid in aerosol sprays by pressurizing

# Secondary 2 Science Exam Papers

**Stuart Donnelly**

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now assume that f is calculated in the command window what is true about the dimensions size of f f will be a column vector f  
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