Periodic Trends Practice Worksheet

Use the periodic table and your knowledge of the periodic trends to answer the following questions.

Identify each element as a metal, metalloid, or nonmetal

a germanium _____

© zinc ____

phosphorous ______

(a) lithium

Which of the two species is larger in each of the following pairs?

(a) N³ or F _____ (b) Mg² or Ca² _____ (c) Fe² or Fe³ _____

(3) Which of the two species is smaller in each of the following pairs?

(a) K" or Li" (b) Au" or Au3" (c) P3 or N3" _____

(i) What atom has the LARGEST atomic radius?

What atom has the SMALLEST atomic radius?

(6) Which is larger?

(a) Mg or Mg²* ____ (b) I or I ____ (c) Cl or Br ____

(d) F or Ne ____ (e) N³ or P³ ____

(7) Put the following in order of increasing radius: Br Se2 Sr2 Kr Rb

(8) Circle the correct answer.

(a) Bromine has a higher / lower ionization energy than Lead

(b) Barium has a higher / lower ionization energy than Radon

© Fluorine has a higher / lower ionization energy than Arsenic

(3) Lithium has a higher / lower ionization energy than Cesium

Which of the two species is larger in each of the following pairs?

(a) Mg or Sr

(b) Be or N (c) P or Cl (d) Ni or Br

Arrange the following in order of increasing electronegativity: S, O, Ne, Al

Periodic Trends Practice Problems With Answers

Marco Cascella

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- 1. Understanding the eBook Periodic Trends Practice Problems With Answers
 - The Rise of Digital Reading Periodic Trends Practice Problems With Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Trends Practice Problems With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Practice Problems With Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Trends Practice Problems With Answers

- Personalized Recommendations
- Periodic Trends Practice Problems With Answers User Reviews and Ratings
- Periodic Trends Practice Problems With Answers and Bestseller Lists
- 5. Accessing Periodic Trends Practice Problems With Answers Free and Paid eBooks
 - Periodic Trends Practice Problems With Answers Public Domain eBooks
 - Periodic Trends Practice Problems With Answers eBook Subscription Services
 - Periodic Trends Practice Problems With Answers Budget-Friendly Options
- 6. Navigating Periodic Trends Practice Problems With Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Trends Practice Problems With Answers Compatibility with Devices
 - Periodic Trends Practice Problems With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Practice Problems With Answers
 - Highlighting and Note-Taking Periodic Trends Practice Problems With Answers
 - Interactive Elements Periodic Trends Practice Problems With Answers
- 8. Staying Engaged with Periodic Trends Practice Problems With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Practice Problems With Answers
- 9. Balancing eBooks and Physical Books Periodic Trends Practice Problems With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Practice Problems With Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Practice Problems With Answers
 - Setting Reading Goals Periodic Trends Practice Problems With Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Practice Problems With Answers

- Fact-Checking eBook Content of Periodic Trends Practice Problems With Answers
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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