

**5.4****THE PERIODIC TABLE: ORGANIZING THE ELEMENTS****SECTION REVIEW****Objectives**

- Describe the origin of the periodic table
- Identify the position of groups, periods, and the transition metals in the periodic table

**Key Terms**

- |                           |                           |               |
|---------------------------|---------------------------|---------------|
| • periodic table          | • metals                  | • nonmetals   |
| • periods                 | • alkali metals           | • halogens    |
| • periodic law            | • alkaline earth metals   | • noble gases |
| • group                   | • transition metals       | • metalloids  |
| • representative elements | • inner transition metals |               |

**Part A Completion**

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

- The periodic table organizes the elements into vertical 1 1. \_\_\_\_\_ and horizontal 2 2. \_\_\_\_\_ in order of increasing 3 3. \_\_\_\_\_. The table is constructed so that elements that have similar chemical properties are in the same 4 4. \_\_\_\_\_. The elements in Groups 1A through 7A are called the 5 5. \_\_\_\_\_. The 6 6. \_\_\_\_\_ make up Group 0. The elements in Groups 2A and 3A are interrupted in periods 4 and 5 by the 7 7. \_\_\_\_\_. and in the periods 6 and 7 by the 8 8. \_\_\_\_\_. The Group 1A elements are called the 9 9. \_\_\_\_\_, and the Group 2A elements are called the 10 10. \_\_\_\_\_. The nonmetals of 11 11. \_\_\_\_\_ are called the halogens. Elements with properties that are intermediate between those of metals and nonmetals are called 12 12. \_\_\_\_\_.

# Periodic Table Section Review Answer Key

**Neil D. Jespersen, Pamela Kerrigan**



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