# The Periodic Table of Elements

In 1869 Dimitri Mendeleev organized the elements in a table based on the properties of the elements.



have the same number of valence electrons.

	th salmos e	2					-	Store 84
1 <sup>st</sup> Period → ⊙	1 H	Greenop 2 à	Group 34	Group 45	- Co		b 6mg 74	2 He
An Additional	Hydrogen	Conchesco .	Spinstern S	Carmetonia P.	(3 united)	Car you soon	CT WHERE	Helium
2nd Period → ③	3 Li	4 Be	5 B	e C	7 N	8	9 F	10 Ne
	Lithium	Becyllium	Boron	Carbon	Nitrogen	Oxygen	Flouring	Neon
3rd Period - (1)	" Na	12 Mg	13 AI	<sup>14</sup> Si	15 P	16 S	17 CI	18 Ar
CJ Stiens)	Sodium	Magnesiam	Aluminom	Silicon	Phosphorus	Swifer	Chlorine	Argon
4th Period - (6)		20 Ca	34 Ga	32 Ge	AS As	34 Se	35 Br	36 Kr
(4 Shells)	Potassium	Calcium	Gallium	Germanium	Personic .			Krypton

[Periods] (rows)

The elements in each period have the same 3 number of shells.

Today, 114 elements are arranged by atomic number in the periodic table. The atomic number is based on the number of in an element.

# **Periodic Table Notes For Middle School**

Eva Brooks, Emma Edstrand, Anders Kalsgaard Møller, Thomas Bjørner **Periodic Table Notes For Middle School:** 

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