Periodic Trends

ATOMIC RADIUS

- What trend in atomic radius do you see as you go down a group/family on the periodic table?
- What causes this trend?
- What trend in atomic radius do you see as you go across a period/row on the periodic table?
- 4. What causes this trend?
- Circle the atom in each pair that has the largest atomic radius.
 - a) Al B

- b) S O c) Br Cl

- d) Na Al e) O F f) Mo Ca
- Put the following elements in order from smallest to largest atomic radius and explain why: C. O. Sn. Sr.

ELECTRONEGATIVITY

- Define electronegativity
- 8. How does the ionic radius of a nonmetal compare with its atomic radius?
- 9. What trend in electronegativity do you see as you go down a group/family on the periodic table?
- 10. What causes this trend?
- 11. What trend in electronegativity do you see as you go across a period/row on the periodic table?
- 12. What causes this trend?
- Circle the atom in each pair that has the greater electronegativity.
- a) Ca Ga b) Li O c) Cl S d) Br As

- e) Ba Sr

GENERAL QUESTIONS

- 14. Which group tends to form +1 ions?
- 15. Which group tends to form +2 ions?
- 16. Which group tends to form -1 ions?
- 17. Which group tends not to form ions or react?

Periodic Trends Ch 5 Answers

Hussin A.Rothana

Periodic Trends Ch 5 Answers:

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Westphal, 2021-09-03 Differentiating Instruction With Menus Chemistry offers teachers everything needed to create a student centered learning environment based on choice This book uses different types of menus that students can use to select exciting advanced level products that they will develop so teachers can assess what has been learned instead of using a traditional worksheet format Topics addressed include chemistry basics measurements atoms chemical bonding and reactions gas laws energy acids and bases and nuclear and organic chemistry Differentiating Instruction With Menus Chemistry contains attractive reproducible menus each based on the levels of Bloom's revised taxonomy as well as incorporating different learning styles. These menus can be used to guide students in making decisions as to which products they will develop after studying a major concept or unit Grades 9 12 Cracking the AP Chemistry Exam, 2019 Edition The Princeton Review, 2018-11-06 Make sure you re studying with the most up to date prep materials Look for The Princeton Review's Cracking the AP Chemistry Exam 2020 ISBN 9780525568186 on sale August 2019 Publisher's Note Products purchased from third party sellers are not quaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product Physical Chemistry for the Biosciences Raymond Chang, 2005-02-11 This book is ideal for use in a one semester introductory course in physical chemistry for students of life sciences The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details Subsequently only basic skills of differential and integral calculus are required for understanding the equations The end of chapter problems have both physiochemical and biological applications

Hospitality and Travel Marketing Alastair M. Morrison,2022-10-27 Fully revised and updated for its fifth edition Hospitality and Travel Marketing provides students with an international and systematic approach to hospitality and travel marketing structured around planning research implementation control and evaluation Written in a user friendly style and structured in a logical and organized manner to aid learning students benefit from the ease of communication practical nature and excellent use of relevant and up to date cases The author's global experience in the industry is emphasized through content on hospitality and travel marketing and other parts of tourism along with a plethora of timely and relevant real life case examples from around the world This new fifth edition is positioned as a post COVID 19 text reflecting the new realities of marketing after the pandemic and has been updated to reflect these current trends in the field including e marketing mobile marketing societal marketing and destination branding It specifically has been updated by Including three new technology chapters on e marketing marketing on social media platforms electronic customer relationship management eCRM and customer co creation in marketing A new chapter on social responsibility societal and social marketing New content on the new realities of the post COVID era and the increasing competitiveness in hospitality and travel greater emphasis on branding disruptive technologies consumer control marketing and generations uses of user generated content and globalization New global case studies throughout with reflective questions to use in class or for self study New

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Periodic Trends Ch 5 Answers:

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