

## Answers

$Q = mc\Delta T$ , where  $Q$  = heat energy,  $m$  = mass, and  $\Delta T$  = change in temp.  
Remember,  $\Delta T = (T_{\text{final}} - T_{\text{initial}})$ . Show all work and proper units.

1. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{1086.75}{15.75(175-25)} = 0.46 \text{ J/g}^\circ\text{C}$$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$Q = mC(T_f - T_i) = 10.0\text{g} (0.90\text{J/g}^\circ\text{C})(55-22) = 297 \text{ J}$$

3. Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{67500 \text{ J}}{(1500 \text{ g})(57-32)} = 1.8 \text{ J/g}^\circ\text{C}$$

4. 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_f - T_i) = 100\text{g}(4.184\text{J/g}^\circ\text{C})(37 - 4) = 14000 \text{ J}$$

5. 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the specific heat capacity of mercury.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{455 \text{ J}}{(25\text{g})(155-25)} = 0.14 \text{ J/g}^\circ\text{C}$$



# Physical Science If8767 Answer Key Heat Calculations

**RD Boyd**



**Physical Science If8767 Answer Key Heat Calculations:**

The book delves into Physical Science If8767 Answer Key Heat Calculations. Physical Science If8767 Answer Key Heat Calculations is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Science If8767 Answer Key Heat Calculations, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Physical Science If8767 Answer Key Heat Calculations
    - Chapter 2: Essential Elements of Physical Science If8767 Answer Key Heat Calculations
    - Chapter 3: Physical Science If8767 Answer Key Heat Calculations in Everyday Life
    - Chapter 4: Physical Science If8767 Answer Key Heat Calculations in Specific Contexts
    - Chapter 5: Conclusion
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  3. In chapter 2, this book will delve into the foundational concepts of Physical Science If8767 Answer Key Heat Calculations. This chapter will elucidate the essential principles that must be understood to grasp Physical Science If8767 Answer Key Heat Calculations in its entirety.
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  6. In chapter 5, this book will draw a conclusion about Physical Science If8767 Answer Key Heat Calculations. The final chapter will summarize the key points that have been discussed throughout the book.
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