

A-Level Chemistry - Year 2

OCR Chemistry A - Paper 1 June - 2019

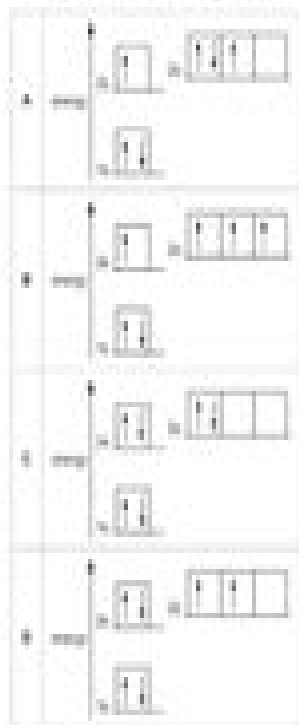
Question - 1

1
SECTION 1

You should spend a maximum of 40 minutes on this section.
Write your answer to each question in the box provided.

Answer all the questions.

- 1 In the diagram below each row represents an orbital and each electron is shown as an arrow.
What happens when the orbital energy of electrons is at each of (a)–(d)?



For answer 1

HOOO



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Oxford Cambridge and RSA

Tuesday 4 June 2019 – Afternoon

A Level Chemistry A

443291 Periodic table, elements and physical chemistry

Time allowed: 1 hour 15 minutes

We need you

- 1 to take the opportunity to showcase your skills and progress

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Please write clearly in black ink. Do not write in the brackets.

Centre number

Candidate number

First name

Last name

INSTRUCTIONS

- 1 Use black ink. You may use an HB pencil for graphs and diagrams.
- 1 Answer all the questions.
- 1 Where appropriate your answers should be supported with writing. Marks may be given for a correct method even if the answer is incorrect.
- 1 Write your answer to each question in the space provided. If additional space is required, see the back papers at the end of this booklet. The question numbers must be clearly shown.

INFORMATION

- The total mark for this paper is 100.
- The marks for each question are shown in brackets ().
- Quality of extended responses will be assessed in questions marked with an asterisk (*).
- This document consists of 10 pages.

Ocr Chemistry Past Papers June 2014

Lei Huang



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Reviewing **Ocr Chemistry Past Papers June 2014**: Unlocking the Spellbinding Force of Linguistics

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