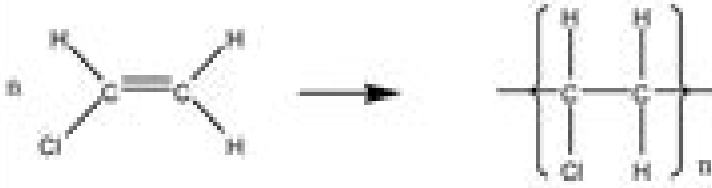


Question	Answer	Mark	Guidance
5 (a)	 <p>Correct polymer with side links ✓</p> <p>Balanced equation for formation of correct polymer - correct use of <math>n</math> in the equation and brackets ✓</p>	2	<p>Displayed formulae <b>MUST</b> be used to award each mark</p> <p><math>n</math> on LHS can be at any height to the left of formula  <b>AND</b> <math>n</math> on the RHS must be a subscript          (essentially below the side link)</p>
(b) (i)	$\text{CH}_2\text{CHCl} + 2\text{O}_2 \rightarrow \text{CO} + \text{CO}_2 + \text{HCl} + \text{H}_2\text{O} \checkmark$	1	<p><b>ALLOW</b> any other correctly balanced equation with the same reactants and products  <b>ALLOW</b> <math>\text{C}_2\text{H}_3\text{Cl}</math> for <math>\text{CH}_2\text{CHCl}</math></p>
(ii)	<p>Sodium hydrogencarbonate neutralises <math>\text{HCl}</math> ✓</p>	1	<p><b>Assume that 'it' refers to sodium hydrogencarbonate but DO NOT ALLOW</b> other chemicals e.g. sodium</p> <p><b>ALLOW</b> <math>\text{NaHCO}_3</math> is a base  <b>ALLOW</b> forms a salt or sodium chloride or <math>\text{NaCl}</math>  <b>ALLOW</b> equation to show formation of <math>\text{NaCl}</math> from <math>\text{NaHCO}_3</math> and <math>\text{HCl}</math> even if not balanced.  <b>IGNORE</b> reacts</p>

# Ocr F322 June 2014 Unofficial Mark Scheme

**Lingsheng Yao**



**Ocr F322 June 2014 Unofficial Mark Scheme:**

## Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Ocr F322 June 2014 Unofficial Mark Scheme**

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