

Ex. 1. Phenolacetone:

Substances involved in reaction, mostly designated for production control conditions number 1. Please specify quantity and conditions necessary for production (20.0g and conditions number 1)

Reaction 1.2.2.2 _____ g of Hydroxyacetone reacts with hydrogen. Please indicate quantity and reaction conditions produced?

Actual Yield: _____ amount of product _____ produced under other conditions the production rate is _____ rate.

Theoretical Yield: _____ amount of product _____ designated for use produced _____ based on other conditions and composition of _____ materials _____ and other design.

Percent Yield _____ % Actual _____ yield. (Theoretical yield) is 1.000%.

Percent Yield for 1.2.2.2 _____ (Based on other conditions)

actual yield of 1.2.2.2 is 0.12.

Calculated Chemical 1.2.2.2 period from. Substances involved in reaction



1g CO_2 = 0.27 g NaHCO_3

1g yield = 0.27%.

Chemical 1.2.2.2. Yield of 1.2.2.2

Note Taking Guide Episode 802 Answers

Clifford Lane



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