

Separation of mixtures

The components of mixtures may be separated based on their physical properties (FF)

Distillation

Example: An industrial plant uses this property to separate mixtures when they contain petroleum and alcohol. This liquid is called

Gasoline

Some techniques used to separate the components of a mixture are:

Filtration: used to separate liquids and solids

Crystallization: used to separate solids

Extraction: used to separate liquids

Evaporation:

<p>Distillation: used to separate liquids with different boiling points</p>	<p>The liquid with the lowest boiling point is vaporized and then condensed to a liquid.</p>
<p>Crystallization: used to separate solids from a liquid solution</p>	<p>The liquid with the lowest boiling point is vaporized and then condensed to a liquid.</p>
<p>Extraction: used to separate liquids with different boiling points</p>	<p>The liquid with the lowest boiling point is vaporized and then condensed to a liquid.</p>

Note Taking Guide Episode 203 Separating Mixtures

KJ Lindholm-Leary



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