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One (ONE) of the graphs below represents the correct relationship between force  $F$  and the electric field  $E$ .



# PHYSICAL SCIENCES

## GRADE 11 PAPER 1



Block P now moves to the left at a **CONSTANT SPEED** of  $2.5 \text{ m/s}$ .

Newton's First Law of Motion in words.



State Newton's Second Law of Motion in words.



# Physical Science Grd11 Exemplar Noember 2014

**M. Elfrink**



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + 2 \text{ H}_2(\text{g}) + \text{O}_2(\text{g})$   $\Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(\text{g}) + \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g})$   $\Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) + 3 \text{ H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + 1 \text{ C}_2\text{H}_6(\text{g})$   $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$   $\Delta h = 1411 \text{ kJ}$  mole  
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$   $\Delta h = 1560 \text{ kJ}$  mole  
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$   $\Delta h = 285 \text{ kJ}$  mole

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web thus for reaction 9 2 in which 1 mole of h<sub>3</sub>o and 1 mole of oh combine to form 2 moles of h<sub>2</sub>o h 55 8 kj as shown note that the enthalpy change h<sub>products</sub> h<sub>reactants</sub> is positive if heat is absorbed that is if h<sub>products</sub>

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