

## Chapter Review (continued)

## Part B. Concept Review

Directions: Answer the following questions on the lines provided.

1. Compare and contrast the following.

a. distance and displacement: Distance how far you when displacement where how many miles or meters have you moved.  
 b. speed and velocity: speed w/ velocity direction and speed.

2. Explain the difference between positive and negative acceleration.

positive acceleration means that you move forward.  
while negative acceleration means going back ward.

3. How are force and motion related?

they are related because exerting force on an object

4. What property of matter is inertia related to?

is the tendency of an object to remain at rest

5. A car travels west for 240 km in 4 h. What is the car's velocity?

$$240/4 = 60 \text{ km/h}$$

6. An object goes from a speed of 9 m/s to a total stop in 3 s. What is the object's acceleration?

$$9(3) = 27$$

7. Explain what causes an unrestrained passenger in a car to fly forward when the car stops suddenly.

because when the car is moving your velocity

Directions: Use Figures 1a and 1b to answer questions 8 and 9.

Figure 1a

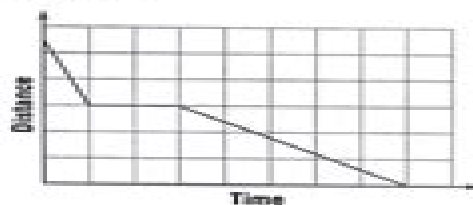
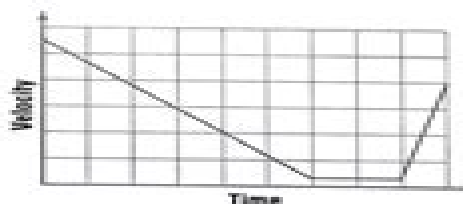


Figure 1b



8. Describe the motion of a car as shown by the distance-time graph in Figure 1a.

negative acceleration + then stop motion.  
the slowing down by negative motion

9. Describe the motion of a biker as shown by the velocity-time graph in Figure 1b.

slowed down + then stop motion + then speed + then acceleration.

# Physical Science Chemical Bonds Part B Concept Review Answers

**CH Cherryholmes**



## Physical Science Chemical Bonds Part B Concept Review Answers:

**Focus on physical science** Charles H. Heimler, 1989      **Addison-Wesley Introduction to Physical Science** Michael B. Leyden, 1988      **Holt Science and Technology 2002** Holt, Rinehart and Winston Staff, 2002      *Physical Chemistry* Peter Atkins, Julio de Paula, 2002 New edition of the overwhelmingly favorite text for the physical chemistry course

*Foundations of Physical Science* Tom Hsu PH.D., 2002      **The Software Encyclopedia**, 1988      **Science Review**, 1966

**Science Books**, 1970      Discover Science: Teacher's annotated edition, 1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today's technological society

**NICEM Index to Nonprint Special Education Materials, Multimedia Learner Volume** National Information Center for Educational Media, National Information Center for Special Education Materials, 1979

**Chemistry and Physics of Energetic Materials** S.N. Bulusu, 2012-12-06 This book represents a collection of lectures presented at the NATO Advanced Study Institute ASI on Chemistry Physics of the Molecular Processes in Energetic Materials held at Hotel Torre Normanna Altavilla Milicia Sicily Italy September 3 to 15 1989 The institute was attended by seventy participants including twenty lecturers drawn from thirteen countries The purpose of the institute was to review the major advances made in recent years in the theoretical and experimental aspects of explosives and propellants In accordance with the format of the NATO ASI it was arranged to have a relatively small number of speakers to present in depth review type lectures emphasizing the basic research aspects of the subject over a two week period Most of the speakers gave two lectures each in excess of one hour with additional time for discussions The scope of the meeting was limited to molecular and spectroscopic studies since the hydrodynamic aspects of detonation and various performance criteria of energetic materials are often covered adequately in other international meetings An attempt was made to have a coherent presentation of various theoretical computational and spectroscopic approaches to help a better understanding of energetic materials from a molecular point of view The progress already made in these areas is such that structure property e.g.

**Femtochemistry: Ultrafast Dynamics Of The Chemical Bond (In 2 Volumes) - Volume 1** Ahmed H Zewail, 1994-09-12 These two volumes on Femtochemistry present a timely contribution to a field central to the understanding of the dynamics of the chemical bond This century has witnessed great strides in time and space resolutions down to the atomic scale providing chemists biologists and physicists with unprecedented opportunities for seeing microscopic structures and dynamics Femtochemistry is concerned with the time resolution of the most elementary motions of atoms during chemical change bond breaking and bond making on the femtosecond  $10^{-15}$  second time scale This atomic scale of time resolution has now reached the ultimate for the chemical bond and as Lord George Porter puts it chemists are near the end of the race against time These two volumes cover the general concepts techniques and applications of femtochemistry Professor Ahmed Zewail who has made the pioneering contributions in this field has from over 250 publications selected the articles for this anthology These

volumes begin with a commentary and a historical chronology of the milestones He then presents a broad perspective of the current state of knowledge in femtochemistry by researchers around the world and discusses possible new directions In the words of a colleague it is a must on the reading list for all of my students all readers will find this to be an informative and valuable overview The introductory articles in Volume I provide reviews for both the non experts as well as for experts in the field This is followed by papers on the basic concepts For applications elementary reactions are studied first and then complex reactions Volume I is complete with studies of solvation dynamics non reactive systems ultrafast electron diffraction and the control of chemical reactions Volume II continues with reaction rates the concept of elementary intramolecular vibrational energy redistribution IVR and the phenomena of rotational coherence which has become a powerful tool for the determination of molecular structure via time resolution The second volume ends with an extensive list of references according to topics based on work by Professor Zewail and his group at Caltech These collected works by Professor Zewail will certainly be indispensable to both experts and beginners in the field The author is known for his clarity and for his creative and systematic contributions These volumes will be of interest and should prove useful to chemists biologists and physicists As noted by Professor J Manz Berlin and Professor A W Castleman Jr Penn State femtochemistry is yielding exciting new discoveries from analysis to control of chemical reactions with applications in many domains of chemistry and related fields e g physical organic and inorganic chemistry surface science molecular biology etc

Russian Chemical Reviews ,1978      Chemistry and Industry Review ,1945      **The Presbyterian** ,1895      *A Complete Preparation for the MCAT.: Knowledge & comprehension of science* James L. Flowers,1988      Bulletin of the Atomic Scientists ,1973-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world      *Engineering* ,1921      **Bulletin of the Atomic Scientists** ,1972-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world      **A Standard Dictionary of the English Language** Isaac Kaufman Funk,Francis Andrew March,1897

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