

(c) Two methods by which rusting of iron can be prevented are and ...

(d) Changes in which only _____ properties of a substance change are called physical changes.

(e) Changes in which new substances are formed are called _____ changes.

Solution:

(a) When carbon dioxide is passed through lime water, it turns milky due to the formation of Calcium carbonate.

(b) The chemical name of baking soda is Sodium hydrogen carbonate.

(c) Two methods by which rusting of iron can be prevented are painting and galvanisation.

(d) Changes in which only physical properties of a substance change are called physical changes.

(e) Changes in which new substances are formed are called chemical changes.

4. When baking soda is mixed with lemon juice, bubbles are formed with the evolution of a gas. What type of change is it? Explain.

Solution:

When baking soda is mixed with lemon juice, bubbles are formed with the evolution of a carbon dioxide gas. This is a chemical change.

Lemon juice + Baking soda → Carbon dioxide + other substances

When a candle burns, both physical and chemical changes take place. Identify these changes. Give another example of a familiar process in which both chemical and physical changes take place.

Sobution:

Candle Burning

Physical change is the melting of the wax.

The chemical change is the burning of gas with the evolution of CO₁

Digestion of Food

Physical change is the breakdown of larger food molecules into simpler ones.

The chemical change is the digestion of food by the action of HCl and enzymes.

6. How would you show that the setting of curd is a chemical change?

Solution:

The setting of curd is a chemical change because curd cannot be turned back to milk. The properties of milk and curd are different.

Physical Science Chapter 6 Answers

Caren Van Slyke

Physical Science Chapter 6 Answers:

Physical Science Chapter 6 Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Physical Science Chapter 6 Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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