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Matr.:

Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermodynamic Energy and Work

Multiple Choice:

1. The electrochemical potential
a. is defined as zero.
b. is arbitrary.
c. is always positive.
d. is always negative.
2. Energy added to a system
a. results mainly in increased translational energy.
b. is lost from the surroundings.
c. is converted to heat.
d. increases the kinetic energy of the system.
3. When a definite reaction occurs in an object,
a. mechanical work is done.
b. energy is lost from the system.
c. there is 100% efficiency.
d. it is not possible to measure heat loss.
4. A slow cyclic process (quasi-statically) is
a. irreversible.
b. reversible.
c. isothermal.
d. isobaric.
5. Compression of a gas is
a. isothermal.
b. isobaric.
c. isochoric.
d. isentropic.
6. Why does liquid water freeze at 0°C?
a. compression.
b. structural rearrangement.
c. transformation.
d. evaporation.
7. Which statement is NOT true about heat?
a. It is always positive.
b. Heat always flows from high to low temperature.
c. Heat always flows from high to low temperature.
d. It is not a state function.
8. Which of the following is a thermal coordinate?
a. temperature.
b. mass.
c. heat.
d. volume.

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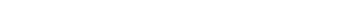
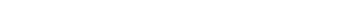
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