

Unit 1 Progress Check: MCQ

1. Which of the following numerical expressions gives the number of moles in 5.0 g of CaO ?

(A) $5.0 \text{ g} \times 56 \text{ g/mol}$

(B) $\frac{5.0 \text{ g}}{56 \text{ g/mol}}$

(C) $\frac{56 \text{ g/mol}}{5.0 \text{ g}}$

(D) $\frac{1}{5.0 \text{ g}} \times \frac{1}{56 \text{ g/mol}}$



2. In a lab a student is given a 21 g sample of pure Cu metal. Which of the following pieces of information is most useful for determining the number of Cu atoms in the sample? Assume that the pressure and temperature in the lab are 1.0 atm and 25°C .

(A) The molar mass of Cu

(B) The density of Cu at 25°C

(C) The volume of the Cu sample

(D) The ratio of the two main isotopes found in pure Cu



3. A 1.0 mol sample of which of the following compounds has the greatest mass?



Novanet Chemistry Answers

Daniel F McAuley



Novanet Chemistry Answers:

This book delves into Novanet Chemistry Answers. Novanet Chemistry Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Novanet Chemistry Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Novanet Chemistry Answers
- Chapter 2: Essential Elements of Novanet Chemistry Answers
- Chapter 3: Novanet Chemistry Answers in Everyday Life
- Chapter 4: Novanet Chemistry Answers in Specific Contexts
- Chapter 5: Conclusion

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