

Blue cards participate _____ of the possible _____ of atoms _____
 1. _____
 2. _____

Atomic fraction number _____ of particles _____ in the particles _____ of an atom _____

Isotopic mass an set of particles _____ $\rightarrow$ an set of particles _____ $\rightarrow$ $\rightarrow$

Isotopes atoms _____ of an element _____ that have different _____ numbers _____ or _____

Isotopes of Hydrogen

Hydrogen-1

1. _____ protons and _____ neutrons

Hydrogen-2

1. _____ protons and _____ neutrons

Hydrogen-3

1. _____ protons and _____ neutrons

Mass fraction mass _____ number of particles _____ and _____ in an atom _____

Example: Carbon-12 _____ 12.011 _____ 12.011 _____ 12.011 _____

Particle	Charge	Mass	Abundance
Proton	+1	1.0073	100%
Neutron	0	1.0087	100%
Electron	-1	0.0005	100%

Atomic mass average _____ of the masses _____ of all the particles in _____

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