

reaction rate:

- the change in \_\_\_\_\_ of the \_\_\_\_\_ per unit time
- speed at which \_\_\_\_\_ disappear and \_\_\_\_\_ appear

The study of reaction rates is called \_\_\_\_\_

- ✓ In order for a reaction to occur, reacting \_\_\_\_\_ must \_\_\_\_\_ effectively.
- ✓ In an \_\_\_\_\_ collision, the \_\_\_\_\_ particles collide with the correct \_\_\_\_\_ and the proper amount of \_\_\_\_\_

Factors Affecting Reaction Rates:

- a. \_\_\_\_\_ of reactants-
  - The \_\_\_\_\_ of the reactants, types of \_\_\_\_\_ involved, and state of \_\_\_\_\_ all affect reaction rates.
- b. \_\_\_\_\_ of reactants-
  - Increasing the concentration of reactants → more \_\_\_\_\_ collisions → increased \_\_\_\_\_ of reaction
  - Increasing the \_\_\_\_\_ of the reactants ( \_\_\_\_\_ only) → increases the \_\_\_\_\_ of reactants → increases \_\_\_\_\_

# Note Taking Guide Episode 1003 Answer Key

**Roman Wölfel**



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