

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M / volume 10.00 mL / acid / base 0.100 M

b. 0.100 M / volume 10.00 mL / acid / base 0.100 M

c. 0.100 M / pH 4.75 / compound that releases acetate
0.100 M / acid / base 0.100 M / acid

Record Values of pH and pH for each and pH for each

0.100 M / 0.100 M / 0.100 M
0.100 M / 0.100 M / 0.100 M

Sample	Acid (pH, Base pH, or Salt pH)	Acids with pH (pH = 4.75)	Conductivity (C or NaC)
HCl	<u>1</u>	<u>1</u>	<u>1</u>
HCl	<u>1</u>	<u>1</u>	<u>1</u>
HCl	<u>1</u>	<u>1</u>	<u>1</u>
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K Payea



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