

2-2

Possible stereoisomers are:

a) **Eis:** **Tirs**, mixture (because of double bond)
 b) isotactic, syndiotactic, atactic (because of -CHCl- groups). However, because of distance between -CHCl- groups tacticity is unlikely to influence ability to crystallize unless rest of groups are all gls or trans.

Use $\begin{bmatrix} 1 \\ 1.51 \end{bmatrix}^2$ for $\begin{matrix} \text{C-O} = a \\ \text{O-C} = b \end{matrix}$ $\begin{matrix} 108^\circ \text{ for C-O-C} = \theta \\ 110^\circ \text{ for C-C-O} = \theta \end{matrix}$

2-3

Graphical solution is easiest using protractor and ruler. Vector summation can be used, also.



If by syndiotactic we mean a succession of **d** and **l** monomer units, the distance turns out to be the same. Another structure is possible with methyls alternating so that the repeat distance would be doubled.

2-4

Refer to Table 2-1, Fig. 2-1.
 Polycyclohexane in cyclohexane:

$$X = 0.136, \quad V_u = 108, \quad v_2 = 0.10$$

$$NV = 2.45 \times 10^3, \quad \text{so that } N = 2.27 \times 10^5 \text{ units/cm}^3$$

Same in solution:

$$\frac{X}{100} = \frac{0.557}{\text{moles in } \frac{1}{2} \text{ original volume}}, \quad V_u = 114, \quad W_u = 2.36 \times 10^{22}, \quad \text{then } v_2 = 0.21$$

2-5

$$v = 0.05, \quad \text{moles/volume} = (\text{g/vol})/(\text{g/mole})$$

If $M_c = M_c/M_r$, per repeat unit = 38 for $\text{C}_{10}\text{H}_{16}\text{O}$ (chain atoms)
 then $M_c = M_c/M_r$, between cross-links = $\frac{5000 \times 38}{5}$ g/mole

$$\rho = 0.913, \quad a = 10^{-8} \text{ meter}$$

$$V = 102 \times 1.21 \times 10^{-2} / 0.800 = 1.58 \times 10^{-7}$$

From Fig. 2-1: $v_2 = 0.059$
 assume additive volumes. Basis: Total volume = 1 cm³

$$1.0 \text{ cm}^3 = V + V' + V_2 = 0.931 + 0.069$$

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Table of Contents Rodriguez Solution Manual

1. Understanding the eBook Rodriguez Solution Manual
 - The Rise of Digital Reading Rodriguez Solution Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Rodriguez Solution Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Rodriguez Solution Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Rodriguez Solution Manual
 - Personalized Recommendations
 - Rodriguez Solution Manual User Reviews and Ratings
 - Rodriguez Solution Manual and Bestseller Lists
5. Accessing Rodriguez Solution Manual Free and Paid eBooks
 - Rodriguez Solution Manual Public Domain eBooks
 - Rodriguez Solution Manual eBook Subscription Services
 - Rodriguez Solution Manual Budget-Friendly Options

6. Navigating Rodriguez Solution Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Rodriguez Solution Manual Compatibility with Devices
 - Rodriguez Solution Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Rodriguez Solution Manual
 - Highlighting and Note-Taking Rodriguez Solution Manual
 - Interactive Elements Rodriguez Solution Manual
8. Staying Engaged with Rodriguez Solution Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Rodriguez Solution Manual
9. Balancing eBooks and Physical Books Rodriguez Solution Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Rodriguez Solution Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Rodriguez Solution Manual
 - Setting Reading Goals Rodriguez Solution Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Rodriguez Solution Manual
 - Fact-Checking eBook Content of Rodriguez Solution Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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