A Periodic Table Logic Problem

Purpose: In this activity you will use a set of clues to identify a set of unknown elements in a periodic table. The elements are the elements with atomic numbers 1-20 and 31-36. A letter of the alphabet is used to represent each unknown element. The letter designation is not related to an element's chemical symbol. Each clue refers to a property of an element or a relationship an element has to other elements in the periodic table. Along with logic and knowledge of properties, you will use the periodic trends to solve the puzzle.

Grade: Neatness and completeness count.

Procedure: Use the following trend clues to place the elements in their proper places on the provided periodic table. Try to identify the elements with direct clues first. For example, B is a direct clue because it identifies only one element. Group 1 Group 2 Group 13 Group 14 Group 15 Group 16 Group 17 Group 18

Work copy

Group I	Group 2		Group 13	Group 14	Group 15	Group 16	Group 17	Group 18
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-	- 1	77	7.7) =	-	2. -	1750	177.51
1-1			-	-	-	7-	-	
1	-		-	1 1 1	Title	- L-	-	-

A - Has one of the highest electronegativities on the table that is NOT a halogen.

B – Has one electron in a 3p orbital.

C – Has five electrons in the 4th energy level.

D – Forms the smallest 2+ ion.

E - Transition Metals are found in these groups

F - Electron configuration is 1s2 2s2 2p6 3s2 3p3

G – The most electronegative element.

H - An ion of this element with a 2+ charge has 18 electrons.

I – This group is known as the Halogens

J — It's shorthand electron configuration must have element M in brackets

Periodic Table Logic Problem Answer Key

M Mosston

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