

The Periodic Table

The Structure of the Periodic Table

- The periodic table displays the elements in order of increasing atomic number i.e. in order of increasing number of protons.
- Elements are arranged in 18 vertical columns called groups that are numbered 1 – 18.
- The seven horizontal rows in periodic table are called periods. The periods are numbered 1-7. The first period contains only two elements, hydrogen and helium.
- Elements within the same groups have similar chemical properties because they have the same number of valence electrons. The number of valence electrons is the same as the group number (in Roman numerals).

PERIODS 1-7		GROUPS 1-18																			
		1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
1		H 1.0079	He 4.0026																		
2		Li 6.941	Be 9.0122											B 10.811	C 12.011	N 14.007	O 15.999	F 18.998	Ne 20.18		
3		Na 22.99	Mg 24.305											Al 26.982	Si 28.086	P 30.974	S 32.06	Cl 35.453	Ar 39.948		
4		K 39.098	Ca 40.078	Sc 44.956	Ti 47.88	V 50.942	Cr 51.996	Mn 54.938	Fe 55.847	Co 58.933	Ni 58.693	Cu 63.546	Zn 65.39	Ga 69.723	Ge 72.64	As 74.922	Se 78.96	Br 79.904	Kr 83.8		
5		Rb 85.468	Sr 87.62	Y 88.906	Zr 91.224	Nb 92.906	Mo 95.94	Tc 97.907	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.6	I 126.9	Xe 131.29		
6		Cs 132.91	Ba 137.33	La 138.91	Hf 178.49	Ta 180.95	W 183.84	Re 186.21	Os 190.23	Ir 192.22	Pt 195.08	Au 196.97	Hg 200.59	Tl 204.38	Pb 207.2	Bi 208.98	Po 209	At 210	Rn 222		
7		Fr 223	Ra 226	Ac 227	Rf 261	Ha 262	Sg 266	Nh 271	Hs 277	Mt 288	Ds 289										

Energy levels and period numbers

The electrons of an atom are arranged in energy levels. The period number tells us in which energy level an atom has its valence electrons.

For example:

Element	In which Group?	In which Period?	How many valence electrons	Which energy level are valence electrons?
Phosphorous	15	3	5	3 rd energy level
Potassium				
Calcium				
Oxygen				

Periodic Table Lessons Fourth Grade

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A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the 'Disha Experts' text.

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Lesson Design for Differentiated Instruction, Grades 4-9 Kathy Tuchman Glass, 2009-01-14 This user friendly resource provides step by step guidance and a detailed template for creating meaningful lessons that are differentiated according to students learning characteristics *Conceptual Chemistry Class XI Vol. I* S K Jain, A book on Conceptual Chemistry **Self-Help to I.C.S.E. Chemistry Class 10 (For 2022-23 Examinations)** Amar Bhutani, RP Gupta K., This book is written strictly in accordance with the latest syllabus prescribed by the Council for the I C S E Examinations in and after 2023 This book includes the Answers to the Questions given in the Textbook Concise Chemistry Class 10 published by Selina Publications Pvt Ltd This book is written by Sunil Manchanda *KVPY - SA : Chemistry for Class 11th by Career Point Kota* Career Point Kota, 2020-07-26 Career Point Kota feel great pleasure to present before you this KVPY SA book Detailed Topic Wise theory supported with example Previous Year Questions Complete Solution This book is designed for the aspirants of KVPY Stream SA As there is no prescribed syllabus for KVPY hence this books is designed considering the topics from where questions have been asked in previous years The book is scientifically structured to prepare aspirants of KVPY Each chapter has detailed topic wise Theory supported with examples to understand the application of concepts followed by Exercise 1 covering the different patterns of questions to give sufficient practice to the students After this Exercise 2 is given covering previous years questions to give exposure to type of questions asked Complete solutions of exercise sheets are also provided in the book itself These solutions are not just sketch rather have been written in such a manner that the students will be able to understand the application of concept and can answer some other related questions too We firmly believe that the book in this form will definitely help a genuine hardworking student We have tried our best to keep errors out of this book Comment and criticism from readers will be highly appreciated and incorporated in the subsequent edition We wish to utilize the opportunity to place on record our special thanks to all team members of Content Development for their efforts to make this wonderful book *Oswal - Gurukul Chemistry Most Likely Question Bank : ICSE Class 10 For 2023 Exam* Oswal - Gurukul, 2022-05-14 **Oswaal Karnataka SSLC Question Bank Class 10 Science Book Chapterwise & Topicwise (For 2024 Exam)** Oswaal Editorial Board, 2023-06-15 Description of the product Latest Board Examination Paper 2023 Held in April 2023 with Board Model Answer Strictly as per the Revised Textbook syllabus blueprint design of the question paper Latest Board specified typologies of questions for exam success Perfect answers with Board Scheme of Valuation Handwritten Topper s Answers for exam oriented preparation KTBS Textbook Questions fully solved Crisp revision with Revision notes and Mind maps Hybrid learning with best in class videos 2 Model Papers solved for Examination Practice 3 Online Model Papers *Oswaal Karnataka SSLC Question Bank Class 10 Science Book Chapterwise & Topicwise (For 2025 Exam)* Oswaal Editorial Board, 2025-01-11 Description of the Product Latest Board Examination Paper 2024 with Board Model Answer Strictly as per the Revised Textbook syllabus blueprint design of the question paper Latest Board specified

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Ansari,Abhishika Kaushik,2021-09-10 1 This book deals with CBSE New Pattern Chemistry for Class 11 2 It is divided into 7 Chapters as per Term 1 Syllabus 3 Quick Revision Notes covering all the Topics of the chapter 4 Carries all types of Multiple Choice Questions MCQs 5 Detailed Explanation for all types of questions 6 3 practice papers based on entire Term 1 Syllabus with OMR Sheet With the introduction of new exam pattern CBSE has introduced 2 Term Examination Policy where Term 1

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