

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is given.

67	La	Lanthanum	138.905	68	Ce	Cerium	140.12	69	Pr	Praseodymium	140.90768	70	Nd	Niobium	144.24	71	Pm	Promethium	144.9127	72	Sm	Samarium	150.36	73	Eu	Europium	151.964	74	Gd	Gadolinium	157.25	75	Tb	Terbium	158.92535	76	Dy	Dysprosium	162.50087	77	Ho	Holmium	164.93032	78	Er	Erbium	167.259	79	Tm	Thulium	168.93032	80	Yb	Ytterbium	173.045	81	Lu	Lutetium	174.967	82	Ac	Actinium	227.0287	83	Th	Thorium	232.0377	84	Pa	Protactinium	231.03688	85	U	Uranium	238.02891	86	Np	Neptunium	237.04817	87	Pu	Plutonium	244.0642	88	Am	Americium	243.06138	89	Cm	Curium	247.07645	90	Bk	Berkelium	247.0703	91	Cf	Californium	251.07958	92	Es	Einsteinium	252.0833	93	Fm	Fermium	257.1037	94	Md	Mendelevium	258.10386	95	No	Nobelium	259.10386	96	Lr	Lutetium	262.10386
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Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

Periodic Table Basic Speedy Study Guides

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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Table Basic Speedy Study Guides**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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