

The Periodic Law

SECTION 2

Select the correct answer to the following questions for the space provided.

1. _____ is the number of protons in the nucleus of an element.
 - (a) something on the periodic table
 - (b) something on Mendeleev's original table
 - (c) **something on Mendeleev's original table**
 - (d) equal to the atomic number
2. _____ Mendeleev stated that elements with similar physical and chemical properties
 - (a) appeared in regular intervals in his periodic table
 - (b) appeared in regular intervals
 - (c) **appeared in regular intervals**
 - (d) were in the same group
3. _____ The modern periodic law states that
 - (a) all elements with the same properties are listed in the same group
 - (b) **the physical and chemical properties of elements are functions of atomic number**
 - (c) elements with the same properties are listed in the same group
 - (d) the chemical properties of elements are functions of atomic number
4. _____ The discovery of the noble gases changed Mendeleev's periodic table.
 - (a) adding a row
 - (b) adding a column
 - (c) **adding a group**
 - (d) adding a period
5. _____ The noble gases have properties of the noble gases is that they are
 - (a) unstable
 - (b) **inert**
 - (c) unstable
 - (d) **highly reactive**
6. _____ Lanthanum, the first element in Group 3, has an atomic number of 57. How many elements in this group have an atomic number of
 - (a) 4
 - (b) 5
 - (c) **6**
 - (d) 3
7. How many elements have an atomic number of 78 and are elements of the
 - (a) **1**
 - (b) **2**
 - (c) **3**
 - (d) **4**
8. How many elements are in this group?
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100. How many elements are in this group?
 - (a) **1**
 - (b) **2**
 - (c) **3**
 - (d) **4**

Periodic Law Chapter 5 Test Answers

EduGorilla Prep Experts

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the EduGorilla Prep Experts text.

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