

## 5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

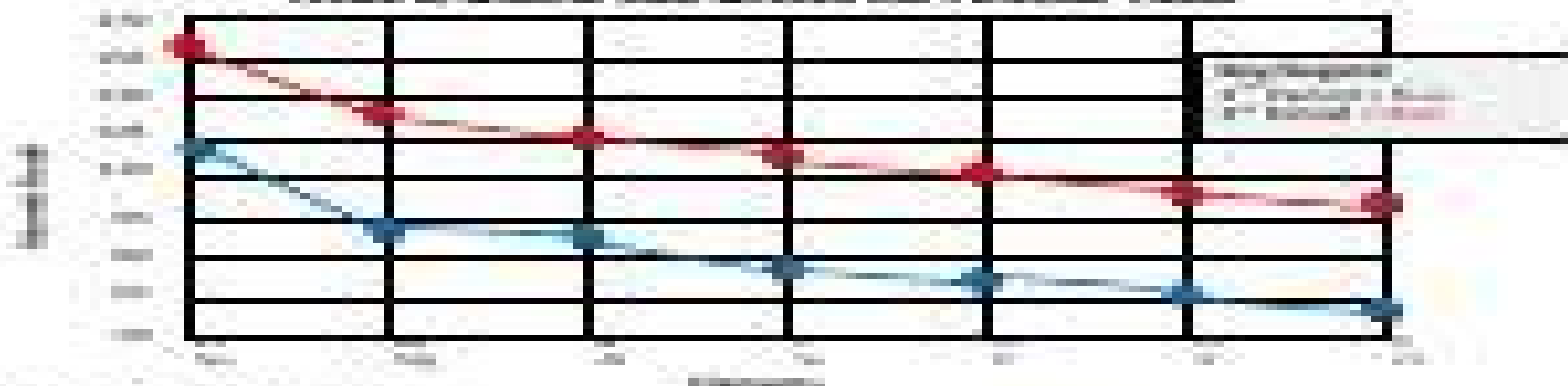
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property stated in the graphing sheet of 5 data points (graph above)
2. Is a slight downward trend in this graph Period? If not, fill in the value graph
3. Label the box as 'not trended' (highlighted)
4. Answer all questions...

|   | 1    | 2    | 3    | 4    | 5    | 6    | 7    | 8    |
|---|------|------|------|------|------|------|------|------|
| 1 | 1H   | 2He  | 3Li  | 4Be  | 5B   | 6C   | 7N   | 8O   |
| 2 | 9Na  | 10Ne | 11Mg | 12Al | 13Si | 14P  | 15S  | 16Cl |
| 3 | 17Ar | 18Kr | 19K  | 20Ca | 21Sc | 22Ti | 23V  | 24Cr |
| 4 | 25Mn | 26Fe | 27Co | 28Ni | 29Cu | 30Zn | 31Ga | 32Ge |
| 5 | 33As | 34Se | 35Br | 36Kr | 37Rb | 38Sr | 39Y  | 40Zr |
| 6 | 41Nb | 42Mo | 43Tc | 44Ru | 45Rh | 46Pd | 47Ag | 48Cd |
| 7 | 49In | 50Sn | 51Sb | 52Te | 53I  | 54Xe | 55Ba | 56La |

Now label the trends in the following table:

Trend in Atomic Size Across the Periodic Table



### Electronegativity Questions

1. Following the trend in atomic size across the periodic table, decreases
2. This means two atoms that are most different properties (electronegativity) they are all different (making electronegativity trend in electronegativity) together

### Electronegativity Questions

1. Electronegativity trend in the periodic table is decreasing (e.g. Na, K, Rb, Cs, Fr) but in electronegativity measurement as you go down is increasing in the periodic table? decreasing
2. This means two atoms that are most different properties (electronegativity) they are all different (making electronegativity trend in electronegativity) together

# Periodicity Activity Okstate Answer Key

**PT Brinkman**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the PT Brinkman text.

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The Journal of the Oklahoma State Medical Association ,1991      *Agrindex* ,1993

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