## Periodic Table Notes: Determining # of Shells and Valence Electrons

#### Directions:

View the Power Point study guide with the same title as this handout. Record all your answers below.

- 1. In the periodic table, each row is called a \_\_\_\_\_\_.
- 2. The elements in each \_\_\_\_\_\_ have the same number of
- 3. Each column in the periodic table is called a \_\_\_\_\_\_.
- 4. Each of the elements in the same \_\_\_\_\_ have the same number of \_\_\_\_ in their outer shells.
- 5. The electrons in the outer shell are called \_\_\_\_\_\_ electrons.
- 6. The \_\_\_\_\_ number equals the \_\_\_\_\_ of \_\_\_\_\_electrons.

Complete the following data table as you view the rest of the slides.

Symbol	Element	Period	# of Shells	Group #	# Valence electrons
С					
Na					
Ne					
н					
Be					
s					
ĸ	1.5				
He					

# **Periodic Table Study Guide Worksheet**

**Albert Kaskel** 

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## **Table of Contents Periodic Table Study Guide Worksheet**

- 1. Understanding the eBook Periodic Table Study Guide Worksheet
  - The Rise of Digital Reading Periodic Table Study Guide Worksheet
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Study Guide Worksheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Study Guide Worksheet
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Study Guide Worksheet
  - Personalized Recommendations
  - Periodic Table Study Guide Worksheet User Reviews and Ratings
  - Periodic Table Study Guide Worksheet and Bestseller Lists
- 5. Accessing Periodic Table Study Guide Worksheet Free and Paid eBooks
  - Periodic Table Study Guide Worksheet Public Domain eBooks
  - Periodic Table Study Guide Worksheet eBook Subscription Services
  - Periodic Table Study Guide Worksheet Budget-Friendly Options

- 6. Navigating Periodic Table Study Guide Worksheet eBook Formats
  - o ePub, PDF, MOBI, and More
  - Periodic Table Study Guide Worksheet Compatibility with Devices
  - Periodic Table Study Guide Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Periodic Table Study Guide Worksheet
  - Highlighting and Note-Taking Periodic Table Study Guide Worksheet
  - Interactive Elements Periodic Table Study Guide Worksheet
- 8. Staying Engaged with Periodic Table Study Guide Worksheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Study Guide Worksheet
- 9. Balancing eBooks and Physical Books Periodic Table Study Guide Worksheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Study Guide Worksheet
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Study Guide Worksheet
  - Setting Reading Goals Periodic Table Study Guide Worksheet
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Study Guide Worksheet
  - Fact-Checking eBook Content of Periodic Table Study Guide Worksheet
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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