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Table of Elements

Use the table to answer the questions.

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Element	Symbol	Number of Protons (Atomic #)	Number of Electrons	Number of Neutrons (most stable isotope)	Atomic Mass (most stable isotope)	Type of Element
Hydrogen	H	1	1	0	1	non-metal
Hellum	He	2	2	2	4	noble gas
Lithium	Li	3	3	- 4	7	alkali metal
Carbon	C	6	6	6	12	non-metal
Oxygen	0	8	8	8	16	non-metal
Sillicon	54	14	14	14	28	metalloid
Chlorine	CI	17	17	18	35	halogen
Calcium	Ca	20	20	20	40	alkaline earth metal
Manganese	Mn	25	25	30	55	transition metal
Iron	Fe	26	26	30	56	transition metal
Lead	Pb	8-2	82	125	207	post-transition metal
Gold	Au	79	79	118	197	transition metal

What is the chemical symbols for iron, lead, and gold?
2. Which element has the fewest protons?
3. In this table, which element has the most neutrons?
4. Which element in this table is a noble gas?
5. Which element has 26 protons?
6. Which elements have 30 neutrons?
7. The atomic number (# of protons) uniquely determines an element. Does the number of neutrons determine an element?
 A proton has a positive charge; an electron has a negative charge. The number of protons and the number of
is the same (this gives the atom a neutral charge).
9. How many protons plus neutrons does chlorine have?
10. The protons and neutrons are located in the nucleus of an atom; tiny electrons orbit around the nucleus. Atomic mass is the weight of the atomic nucleus; it is the number of protons plus the number of

Review Elements And Their Properties Answers Key

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Table of Contents Review Elements And Their Properties Answers Key

- 1. Understanding the eBook Review Elements And Their Properties Answers Key
 - The Rise of Digital Reading Review Elements And Their Properties Answers Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Review Elements And Their Properties Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Elements And Their Properties Answers Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Review Elements And Their Properties Answers Key
 - Personalized Recommendations
 - Review Elements And Their Properties Answers Key User Reviews and Ratings
 - Review Elements And Their Properties Answers Key and Bestseller Lists
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 - Highlighting and Note-Taking Review Elements And Their Properties Answers Key
 - Interactive Elements Review Elements And Their Properties Answers Key
- 8. Staying Engaged with Review Elements And Their Properties Answers Key

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- Participating in Virtual Book Clubs
- Following Authors and Publishers Review Elements And Their Properties Answers Key
- 9. Balancing eBooks and Physical Books Review Elements And Their Properties Answers Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Review Elements And Their Properties Answers Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Elements And Their Properties Answers Key
 - Setting Reading Goals Review Elements And Their Properties Answers Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Elements And Their Properties Answers Key
 - Fact-Checking eBook Content of Review Elements And Their Properties Answers Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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