

10.4 Electrophilic Reactions

General Introduction An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

Electrophilic Substitution

Electrophilic Substitution reactions are common in the chemistry of alkenes and are of two types:

- Electrophilic addition to form a saturated product (as in the bromination of alkenes).
- The substitution reaction of H_2O and H^+ to form H_2O^+ , which then reacts with an alkene to form a carbocation.

Electrophilic Substitution reactions are also common in the chemistry of alkenes.

- The electrophile attacks the double bond, forming a carbocation intermediate.
- The carbocation intermediate then reacts with a nucleophile to form the final product.

When reacting alkenes with H^+ to form the following products:

Electrophilic Substitution

- In the following mechanism of alkenes reacting with H^+ to form a carbocation, the electrophile attacks the double bond and a carbocation is formed. _____
- Write down the mechanism for the reaction of alkenes with H^+ to form a carbocation. _____
- Explain, using curly arrows, why the carbocation is formed at the more substituted carbon. _____
- Write down the mechanism for the reaction of alkenes with H^+ to form a carbocation. _____

- Write down the mechanism for the reaction of alkenes with H^+ to form a carbocation. _____

Electrophilic Substitution

- Write down the mechanism for the reaction of alkenes with H^+ to form a carbocation. _____

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Michael Brown



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