ATOMIC STRUCTURE AND THE PERIODIC TABLE

Given the following isotopes, determine the atomic number, the mass number, the number electrons and neutrons.

Symbol 1311	Atomic Number	Mass Number	Protons	Electrons	Neutrons Isotope		
35S	53				- cuirons	Isotope Name	
	16	131	53	53	78	iodine -13	
2He		35	16	16	19		
27AI	2	4	2	2		Julphur - 3	
81Kr	13	27	13	12	2	helium - 4	
	36	81		15	14	alvarinen -	
81 37Rb			36	36	45	Krypten -8	
	37	13	37	37	44	abdrum-	

PARTA

Complete the following chart by writing the symbol for the isotope of the following elements. In addition, give the number of protons, electrons, mass number and atomic number and complete the

Element	Neutrons	Protons	Electrons	Mass	Atomic	Isotope
Uranium-	145	92	92	Number 237	92	Symbol 23113
Chlorine-	28	17	17	45	17	4501
Oxygen-	9	7	*	17	X	1367
Boron-	6	5	5	11	5	"R
Beryllium-	5	4	4	9	11	1 PR.
Hydrogen-	1	1	1	2	1	214
Carbon-	8	6	4	14	6	14C

Physical Science Chapter 4 Answers

Hans U. Fuchs, Federico Corni

Physical Science Chapter 4 Answers:

The book delves into Physical Science Chapter 4 Answers. Physical Science Chapter 4 Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Physical Science Chapter 4 Answers, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Physical Science Chapter 4 Answers
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