

Physical Science
Chapter 2

Properties of Matter

Classifying Matter

Substances

- **Element:** a substance that can't be separated down into simpler substances.
- **Atom:** the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

2 Types of Mixtures

1. **Heterogeneous Mixture:** have materials that are not evenly distributed and noticeably different from one another.
 2. **Homogeneous Mixture:** have materials that are evenly distributed so it is difficult to distinguish one substance from another.
 - a. Also called a solution
 - b. Remain constantly and uniformly mixed
- Which type of mixture is milk?
 - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

Solutions

Solution: when substances dissolve and form a homogenous mixture.

- Examples: windshield wiper fluid, sea water, lemonade

Colloids

Colloid: contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
 - What happens to the size of the drops of fat in milk when milk is homogenized?

The Scattering of Light

- Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

Physical Science Chapter 2 Review Answers Properties Of Matter

Walter George Whitman



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Holt Physical Science Mapi M. Cuevas, 1994

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