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Date: _____

Class: Physical Sciences

Physics 101: Chapter 10 Quiz: Thermal Energy and Heat

Multiple Choice

1. The microscopic quantity associated with heat is:
a) temperature.
b) internal energy.
c) thermal energy.
d) heat flow.
2. Energy added to a system:
a) results in a temperature increase.
b) always leads to a temperature increase.
c) is proportional to the change in temperature.
d) increases the kinetic energy of the system.
3. When a definite quantity of heat is added to an object:
a) temperature will be lower.
b) temperature is not due to motion.
c) there will be 100% efficiency.
d) it gains energy from the heat source.
4. A direct relation between temperature and heat is:
a) proportional.
b) exponential.
c) exponential decay.
d) exponential growth.
5. Temperature differences between two objects:
a) increase.
b) decrease.
c) approach thermal equilibrium.
d) remain constant.
6. Heat flow from a hot object to a cold object is:
a) impossible.
b) spontaneous.
c) reversible.
d) inefficient.
7. Which statement is NOT true about heat?
a) It is energy in transit.
b) It is the transfer of energy from one object to another.
c) It is the transfer of energy from one object to another.
d) It is the transfer of energy from one object to another.
8. Which of the following is a thermal equilibrium?
a) water.
b) air.
c) ice.
d) steam.

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David Baud



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