#### Chapter 6 Chemical Bonds

# Section 6.2 Covalent Bonding (pages 165-169)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

#### Reading Strategy (page 165)

Relating Text and Visuals As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-fitting	
Electron cloud	

#### Covalent Bonds (pages 165-167)

- Describe a covalent bond. \_\_\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - a. electron dot b. structural formula
  - c. space-filling d. electron cloud
- 3. Describe a molecule. \_\_\_\_\_
- 4. Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus.
- 5. Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - a. zero b. one
  - c. two d. three

# Physical Science Chapter 6 Chemical Bonds Chapter Review

**Sabine Zange** 

**Physical Science Chapter 6 Chemical Bonds Chapter Review:** 

Enjoying the Track of Term: An Psychological Symphony within **Physical Science Chapter 6 Chemical Bonds Chapter Review** 

In a world consumed by displays and the ceaseless chatter of instantaneous transmission, the melodic beauty and mental symphony developed by the prepared word usually fade into the back ground, eclipsed by the relentless noise and disruptions that permeate our lives. Nevertheless, nestled within the pages of **Physical Science Chapter 6 Chemical Bonds Chapter Review** a wonderful literary value brimming with natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by an outstanding composer of language, this charming masterpiece conducts readers on a mental journey, well unraveling the hidden melodies and profound affect resonating within each cautiously constructed phrase. Within the depths of the emotional assessment, we can discover the book is key harmonies, analyze their enthralling writing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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