

Chemistry 12 Worksheet 2-2 LeChatelier's Principle

Unit 2 - Chemical Equilibrium

KEY

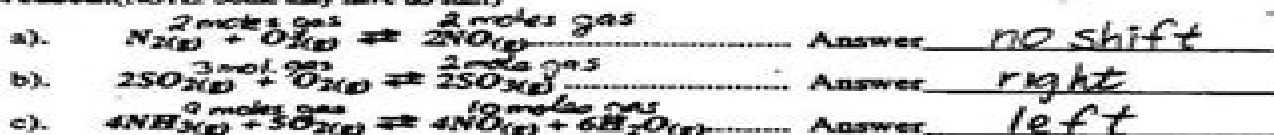
KEY

TOTAL = 70 marks.

1. In order to decide what effect a change in total pressure will have on an equilibrium system with gases, what is the first thing you should do when given the balanced equation?

- Add up the moles of gas on each side of the equation.

2. Predict which way the following equilibrium systems will shift when the total pressure is increased. (NOTE: Some may have no shift)



3. Which way will the following equilibrium shift if the total pressure on the system is decreased?



It will shift right (more moles) to counteract the decrease in pressure.

Answer right

4. Explain why a flask filled with $NO_2(g)$ and $N_2O_4(g)$ will get darker when heated. Use the equation:



- heating will cause the equil^m to shift right.
This will produce more NO_2 which is dark brown. Hence the colour will get darker.

5. State Le Chatelier's Principle.

When a stress is applied to a system at equilibrium, the equil^m will shift (left or right) so as to counteract the imposed stress.

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LP Steffe



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Water-Insoluble Drug Formulation Ron Liu, 2018-03-12 Properties and Formulation From Theory to Real World Application Scientists have attributed more than 40 percent of the failures in new drug development to poor biopharmaceutical properties particularly water insolubility Issues surrounding water insolubility can postpone or completely derail important new drug development Even the much needed reformulation of currently marketed products can be significantly affected by these challenges More recently it was reported that the percentage increased to 90% for the candidates of new chemical entities in the discovery stage and 75% for compounds under development In the most comprehensive resource on the topic this third edition of Water Insoluble Drug Formulation brings together a distinguished team of experts to provide the scientific background and step by step guidance needed to deal with solubility issues in drug development Twenty three chapters systematically describe the detailed discussion on solubility theories solubility prediction models the aspects of preformulation biopharmaceutics pharmacokinetics regulatory and discovery support of water insoluble drugs to various techniques used in developing delivery systems for water insoluble drugs This book includes more

than 15 water insoluble drug delivery systems or technologies illustrated with case studies and featuring oral and parenteral applications Highlighting the most current information and data available this seminal volume reflects the significant progress that has been made in nearly all aspects of this field The aim of this book is to provide a handy reference for pharmaceutical scientists in the handling of formulation issues related to water insoluble drugs In addition this book may be useful to pharmacy and chemistry undergraduate students and pharmaceutical and biopharmaceutical graduate students to enhance their knowledge in the techniques of drug solubilization and dissolution enhancement *Children's Books in Print*, 2007, 2006 *ASHRAE Transactions* American Society of Heating, Refrigerating and Air-Conditioning Engineers, 1994

Assessment that Informs Practice, 2000 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science

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Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02

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Synthesis Green Metrics John Andraos, 2018-12-07 Green chemistry promotes improved syntheses as an intellectual endeavour that can have a great impact both on preserving and utilizing our planet's finite resources and the quality of human life This masterful accomplishment provides an evaluation of environmental impact metrics according to life cycle assessment analysis based on the Mackay compartment environmental model and Guine environmental impact potentials formalism Assumptions limitations and dealing with

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