

Name _____ Date _____ Class _____

17

THERMOCHEMISTRY

SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

► Energy Transformations (page 505)

- What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry
- Where the use of energy is concerned (in a scientific sense), when is work done?
Work is done when a force is used to move an object.
- Circle the letter next to each sentence that is true about energy:
 - ☒ Energy is the capacity for doing work or supplying heat.
 - ☒ Energy is detected only because of its effects.
 - ☐ Heat is energy that transfers from one object to another because they are at the same temperature.
 - ☒ Gasoline contains a significant amount of chemical potential energy.
- Circle the letter next to each sentence that is true about heat.
 - ☒ One effect of adding heat to a substance is an increase in the temperature of that substance.
 - ☐ Heat always flows from a cooler object to a warmer object.
 - ☒ If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

► Exothermic and Endothermic Processes (pages 506–507)

- What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system.

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Raffaela Di Napoli



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Prentice Hall Chemistry ,2000 Prentice Hall Science Explorer Michael J. Padilla,2002 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Children's Books in Print, 2007** ,2006 Norman Hall's Asvab Preparation Book Norman Hall,2015-01-02 Everything you need to know thoroughly covered in one book five ASVAB practice tests answer keys tips to boost scores military enlistment information study aids **Industrial Chemist and Chemical Manufacturer** ,1929

The Industrial Chemist and Chemical Manufacturer ,1929 Education Outlook ,1902 **Principles of Chemical Kinetics** James E. House,2007-08-30 James House s revised Principles of Chemical Kinetics provides a clear and logical description of chemical kinetics in a manner unlike any other book of its kind Clearly written with detailed derivations the text allows students to move rapidly from theoretical concepts of rates of reaction to concrete applications Unlike other texts House presents a balanced treatment of kinetic reactions in gas solution and solid states The entire text has been revised and includes many new sections and an additional chapter on applications of kinetics The topics covered include quantitative relationships between molecular structure and chemical activity organic inorganic chemistry biochemical kinetics surface kinetics and reaction mechanisms Chapters also include new problems with answers to selected questions to test the reader s understanding of each area A solutions manual with answers to all questions is available for instructors A useful text for both students and interested readers alike Dr House has once again written a comprehensive text simply explaining an otherwise complicated subject Provides an introduction to all the major areas of kinetics and demonstrates the use of these concepts in real life applications Detailed derivations of formula are shown to help students with a limited background in mathematics Presents a balanced treatment of kinetics of reactions in gas phase solutions and solids Solutions manual available for instructors *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall,Michael J. Padilla,2004-10 Clinical Biochemistry John S. Varcoe,2001 Clinical biochemistry is an analytical and interpretative science The analytical part involves the determination of the level of chemical components in body fluids and tissues The interpretative part examines these results and uses them in the diagnosis of disease the screening for susceptibility to specific diseases and the monitoring of the progress of treatment This book is designed to cover the major techniques and analytical instruments used in clinical biochemistry Each chapter of this book is based on a specific technique or techniques with associated instrumentation These are discussed in some detail A historical introduction is included for most of the techniques and the current uses of the techniques are presented Following that is a series of practical exercises The first

exercises in most of the chapters are a general introduction to the technique leading to those with a clinical bias Where applicable the clinical practical exercises are associated with a case history and or the discussion of the relevance of the assay to diagnosis and prognosis and to the monitoring of recovery Each chapter concludes with a selection of appropriate references

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