

Your instructor may ask you to repeat one of your trials if your error is greater than 5%.

INTERPRETATION OF DATA

Your data will be used to calculate the theoretical volume of the air in the experiment during the various temperature changes using Charles' Law. The volume of hot air is measured directly by filling the flask with water up to the wax pencil mark and measuring the volume of water (#4, data sheet). The volume of the air after cooling is found by subtracting the volume of water remaining in the cooled flask (#3, data sheet) from the volume held by the full flask (#4). The volume of water sucked into the flask upon cooling represents the extent of contraction of the hot air upon cooling.

Charles' Law can be expressed as:

$$V_{\text{hot gas}} = \text{constant} \times T_{\text{hot gas}} ; \quad V_{\text{cold gas}} = \text{constant} \times T_{\text{cold gas}}$$

or

$$\frac{V_{\text{cold}}}{V_{\text{hot}}} = \frac{T_{\text{cold}}}{T_{\text{hot}}}$$

The temperatures used must be on the Kelvin temperature scale ($K = ^\circ\text{C} + 273.15$). Since your temperature readings have a precision of $\pm 0.1^\circ\text{C}$, add 273.15 but then round to one decimal place. Blanks have been provided on the data sheet for entering these values. From the formula shown above,

$$V_{\text{cold}} = \left(\frac{T_{\text{cold}}}{T_{\text{hot}}} \right) V_{\text{hot}}$$

Use this expression to calculate what the volume of cold air should be.

Calculate the percentage difference between the calculated and measured values as follows:

$$\% \text{ difference} = \frac{(\text{calculated volume of cold air} - \text{measured volume of cold air})}{(\text{calculated volume of cold air})} \times 100 \%$$

Your prediction for Absolute Zero can be found by plotting the volume of gas vs. temperature and extrapolating to zero volume (remember that the molecules of an ideal gas occupy no space). The temperature at zero volume is absolute zero.

To do this, prepare a computer-generated scatter plot of the data with temperature on the y-axis and volume on the x-axis. There will be four data points. Do not connect the dots.

Data point	Volume, mL	Temperature, $^\circ\text{C}$
1	hot air, all trials averaged (#5)	boiling water, all trials averaged (#1)
2	cold air, trial 1 (#6)	ice bath, trial 1 (#2)
3	cold air, trial 2 (#6)	tap water, trial 2 (#2)
4	cold air, trial 3 (#6)	warm water, trial 3 (#2)

Have the program add a best fit line through the data and display the equation of this line. The y-intercept of this equation is your prediction of Absolute Zero from your data. An example follows.

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Lingsheng Yao



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Table of Contents Prentice Hall Chemistry 110 Lab Manual Answer Key

1. Understanding the eBook Prentice Hall Chemistry 110 Lab Manual Answer Key
 - The Rise of Digital Reading Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry 110 Lab Manual Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Personalized Recommendations
 - Prentice Hall Chemistry 110 Lab Manual Answer Key User Reviews and Ratings
 - Prentice Hall Chemistry 110 Lab Manual Answer Key and Bestseller Lists
5. Accessing Prentice Hall Chemistry 110 Lab Manual Answer Key Free and Paid eBooks
 - Prentice Hall Chemistry 110 Lab Manual Answer Key Public Domain eBooks
 - Prentice Hall Chemistry 110 Lab Manual Answer Key eBook Subscription Services
 - Prentice Hall Chemistry 110 Lab Manual Answer Key Budget-Friendly Options

6. Navigating Prentice Hall Chemistry 110 Lab Manual Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Prentice Hall Chemistry 110 Lab Manual Answer Key Compatibility with Devices
 - Prentice Hall Chemistry 110 Lab Manual Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Highlighting and Note-Taking Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Interactive Elements Prentice Hall Chemistry 110 Lab Manual Answer Key
8. Staying Engaged with Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry 110 Lab Manual Answer Key
9. Balancing eBooks and Physical Books Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry 110 Lab Manual Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Setting Reading Goals Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Fact-Checking eBook Content of Prentice Hall Chemistry 110 Lab Manual Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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