

11-3 Practice Problems

1. Identify the limiting reactant when 1.22-g of O_2 reacts with 1.65-g of H_2 to produce water.

2. Identify the limiting reactant when 4.68-g of Fe reacts with 2.88-g of S to produce FeS .

3. Identify the limiting reactant when 2.87-g of $\text{Mg}(\text{OH})_2$ reacts with 12.84-g of HCl to form MgCl_2 and water.

4. Identify the limiting reactant when 6.25-g of AgNO_3 reacts with 4.12-g of NaCl to form NaNO_3 and AgCl .

5. Identify the limiting reactant when 7.81-g of HCl reacts with 5.24-g of NaOH to produce NaCl and H_2O .

6. Identify the limiting reactant when 6.33-g of H_2SO_4 reacts with 5.92-g of NaOH to produce Na_2SO_4 and water.

7. Identify the limiting reactant when 49.25-g of CaC_2 reacts with 33.71-g of water to produce $\text{Ca}(\text{OH})_2$ and C_2H_2 .

8. Identify the limiting reactant when 63.14-g of CaCl_2 reacts with 74.68-g of Na_2CO_3 to produce CaCO_3 and NaCl .

9. Identify the limiting reactant when 4.687-g of SF_4 reacts with 6.281-g of I_2O_5 to produce IF_3 and SO_2 .

10. If 4.1-g of Cr is heated with 9.3-g of Cl_2 , what mass CrCl_3 will be produced?

11. What mass of SO_3 is produced from the reaction between 31.5-g of S_8 and 8.65-g of O_2 ?

12. What mass of SO_3 is produced from the reaction of 12.4-g of SO_2 and 3.45-g of O_2 ?

13. What mass of H_2SO_4 is produced from the reaction of 6.58-g of SO_3 and 1.64-g of H_2O ?

14. What mass of CdS is produced if 8.47-g of cadmium reacts with 2.51-g of Sulfur?

Prentice Hall Chemistry Practice Problem

Christian G. Meyer



Prentice Hall Chemistry Practice Problem:

Whispering the Secrets of Language: An Mental Journey through **Prentice Hall Chemistry Practice Problem**

In a digitally-driven world where displays reign supreme and quick connection drowns out the subtleties of language, the profound techniques and psychological subtleties hidden within phrases usually go unheard. Yet, nestled within the pages of **Prentice Hall Chemistry Practice Problem** a captivating literary value sporting with fresh feelings, lies a fantastic journey waiting to be undertaken. Written by a skilled wordsmith, that charming opus attracts visitors on an introspective journey, gently unraveling the veiled truths and profound impact resonating within the very fabric of each and every word. Within the mental depths of the poignant evaluation, we can embark upon a honest exploration of the book is core styles, dissect its interesting writing type, and yield to the effective resonance it evokes deep within the recesses of readers hearts.

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