

Chapter 6 Chemistry Quiz

1. The Lewis dot symbol consists of the symbol for the element surrounded by dot(s). What does the dot or dots represent?

electron configuration

valence electrons

atomic number

atomic mass

core electrons

2. Which of the following contains ionic bonding?

CO

SrF₂

Al

OCI₂

HCl

3. How many electrons are in a single bond?

0

1

2

4

6

4. How many electrons are in a triple bond?

0

1

2

4

6

5. Select the strongest bond in the following group.

C-S

C-O

C=C

C≡N

C-F

6. Which is the correct order for increasing bond strength? (1) C≡C (2) C=C (3) C-C

(3) < (2) < (1)

(3) < (1) < (2)

(1) < (2) < (3)

Prentice Hall Chemistry Chapter 6 Quiz

Wenbin Ji



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Decoding **Prentice Hall Chemistry Chapter 6 Quiz**: Revealing the Captivating Potential of Verbal Expression

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