### **Predicting Products of Chemical Reactions**

This worksheet is designed to help you predict products of simple reactions of the four basic reaction types (synthesis, decomposition, single replacement, and double replacement) and combustion reactions.

For the first few reactions, the type of reaction is listed, you should predict the products, then balance. Further questions just have the reactions listed and you should decide on the type of reaction, as well as the correct products. Many of these reactions fall into the category of redox reactions, though do not let that confuse you...each can be described in terms of the four basic reaction types (except the combustion reactions).

Although states (s, l, g, aq) of the reactants and products are very important in a chemical reaction, don't worry about determining those for these problems. Rather, focus on what products might result from the reactants given. Pay particular attention to the ionic charge of species that you know form ions with only one possible charge (s.g., alkali metals, alkaline earth metals, halogens, etc.)

a. Combustion:	$C_{\pm}H_{12}$		+	$O_2$	-	
b. Combustion:	$C_aH_a$		+ 2	$O_2$	-	
c. Combustion:	$C_0H_{00}O$		+	$O_2$	-	
1. Synthesis:		мд	+	12	-	
2. Double displacement:		CuCl <sub>2</sub>	+	$H_0S$		
3. Double displacement:		NaOH:	+	HClO <sub>4</sub>	-	
4. Decomposition:		$ZnCO_3$	+	heat	-	
5. Single replacement:		HCI	+	Zo	-	
6		Na	+	MgCl <sub>2</sub>	-	
7		CaCl <sub>2</sub>	+	$K_2CO_3$	-	
8		K	+	Cl <sub>2</sub>	-	
9		BaCl <sub>2</sub>	+	$K_0PO_4$		
10		$H_2SO_{\alpha}$	+	KOH	-	
11		Alg(CO	a)bi	+:	heat	-
12		Alt	# 9	$O_2$	-	
13		Pb(NO <sub>3</sub>	) <sub>2</sub>	+	KOH	-
14.		$H_2SO_{\pi}$	+	BaCl <sub>2</sub>	-	
15		Ca	+	AgCl	-	
16		$H_0 P O_{\alpha}$	+	FeBry	-	
17		1.4	+	N <sub>2</sub>		
18		HCI	+	Mg(OH	) <sub>2</sub>	-
19		Mg(OH	02	+	heat	-
20.		Fe(OH)	9	+	heat	

# **Predicting Products Of Chemical Reactions Easy Practice**

**Clemens Wendtner** 

**Predicting Products Of Chemical Reactions Easy Practice:** 

Unveiling the Magic of Words: A Overview of "Predicting Products Of Chemical Reactions Easy Practice"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually aweinspiring. Enter the realm of "**Predicting Products Of Chemical Reactions Easy Practice**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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