#### Predicting Products of Chemical Reactions

This worksheet is designed to help you predict products of simple reactions of the four basic reaction types (synthesis, decomposition, single replacement, and double replacement) and combustion reactions.

For the first few reactions, the type of reaction is listed, you should predict the products, then balance. Further questions just have the reactions is listed and you should decide on the type of reaction, as well as the correct products. Many of these reactions fall into the category of redox reactions, though do not let that confuse you...each can be described in terms of the four basic reaction types (except the combustion reactions).

Although states (s, l, g, aq) of the reactants and products are very important in a chemical reaction, don't werry about determining those for these problems. Rather, focus on what products might result from the reactants given. Pay particular attention to the ionic charge of species that you know form ions with only one possible charge (e.g., alkali metals, alkaline earth metals, halogens, etc.)

20	Combustion:	$C_6H_{12}$	+-	9 0,		6002 + 6HO
2	Combustion:	2.C <sub>4</sub> H <sub>4</sub>	+	11 02	1000	8 CO2 + 6420
3.	Combustion:	$C_0H_{10}O$	,	+ "7	Oz	→ 600 + 5H+0
4.	Synthesis:	Mg	+	I.a	*	-MaI:
35.	Double displacement:	CuCl <sub>2</sub>	+	$H_1S$	$\rightarrow$	CuS + 2HCI
- 6.	Double displacement.	NaOH		HCIO4	>	NaClOy + HEO
7.	Decomposition:	$Z_0CO_3$		heat	-	Zn0 + Co.
(8)	Single replacement:	2.HCI		Zn		ZnCl. + Hz
9.	sing le	2Na	+	$MgCl_2$	-	2 Nacl . Mg
10.	double	CaCl <sub>2</sub>	*	$K_2CO_3$		Caco, · 2KCI
11.	synthesis	<b>2</b> .K	+	$CII_{\ell}$	<del></del>	2KC1
12.	double	3 BaCl,	+	2 K <sub>3</sub> PO <sub>4</sub>	-	Bas (PON) + GKC1
13.	double	H <sub>2</sub> SO <sub>4</sub>	+	2 кон	-	K2 SO4 + 2H2O
14,	Synthesis	4 AL	+	3 O2		2 A1203
(15)	_domble	Pb(NO <sub>3</sub>	) <sub>2</sub>	+ 2	кон	→ Pb(OH), + 2KNO3
16.	double	H <sub>2</sub> SO <sub>4</sub>	+	BaCl <sub>2</sub>		2 HC1 + B & SO4
(17)	single	Ca	+	2 AgCI	-+	Cacle + 2Ag
18.	double	$H_bPO_4$		$\mathrm{FeB}\tau_{\mu}$	-	3 HBr + Fe PD4
(19.)	synthesis .	6 Li		MrNo		2 L : 3 N
20.	double	2 HCI	+	Mg(OH)	) <sub>1</sub>	→ MgCl2 + 2H20

# **Predicting Products Of Chemical Equations Answers**

**Tao Wei** 

**Predicting Products Of Chemical Equations Answers:** 

Predicting Products Of Chemical Equations Answers: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have charmed audiences this year. Predicting Products Of Chemical Equations Answers: Colleen Hoovers "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Predicting Products Of Chemical Equations Answers: Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic: Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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