

1. According to the Bronsted—Lowry definition, which chemical species can function *both* as an acid and as a base?

- a. Cl^-
- b. SO_4^{2-}
- c. NH_4^+
- d. HCO_3^-
- e. H_3O^+

2. In the reaction $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$ which is an acid—base conjugate pair?

- a. H_2O and HCN
- b. H_2O and OH^-
- c. CN^- and H_2O
- d. HCN and OH^-

3. Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX ?

- a. It is less basic than a 1 M solution of NaY .
- b. It is more basic than a 1 M solution of NaY .
- c. It yields a neutral solution.
- d. It is more concentrated than a 1 M solution of NaY .

4. The pH of a 0.03 M HCl solution is

- a. 1.5
- b. 2.5
- c. 3.5
- d. 12.5

5. The pH of a 1.0×10^{-2} M $\text{Ba}(\text{OH})_2$ solution at 25°C is

- a. 2.7
- b. 3.0
- c. 11.0
- d. 11.3

6. Which is the strongest acid?

- a. HClO
- b. HClO_2
- c. HClO_3
- d. HClO_4

7. Which salt reacts with water (hydrolyzes) to produce a basic solution?

- a. $\text{NaC}_2\text{H}_3\text{O}_2$
- b. NaNO_3
- c. NH_4Cl
- d. BaSO_4

8. The *weakest* of the bases listed is

Acid	Conjugate Base	K_a (Ionization Constant of Acid)
HCl	Cl^-	100% ionized
HSO_4^-	SO_4^{2-}	1.2×10^{-2}
H_2S	HS^-	5.7×10^{-8}
HS^-	S^{2-}	1.2×10^{-13}

- a. Cl^-
- b. CN^-
- c. HS^-
- d. S^{2-}
- e. SO_4^{2-}

9. The oxide of which element will react with water to form the strongest acid?

	Main Groups							
Period	I	II	III	IV	V	VI	VII	O
First								
Second	X	Y	Z	P		Q	S	U
Third	W					R	T	M

- a. W
- b. M
- c. P
- d. R
- e. Z

10. Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate $\text{KC}_2\text{H}_3\text{O}_2$ is less alkaline than a 0.10 M solution of potassium cyanide, KCN ?

- a. Hydrocyanic acid is a weaker acid than acetic acid.
- b. Hydrocyanic acid is less soluble in water than acetic acid.
- c. Cyanides are less soluble than acetates.
- d. Acetic acid is a weaker acid than hydrocyanic acid.

11. In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH , the properties of the solution at the equivalence point will correspond exactly to the properties of

- a. a 0.100 M sodium solution.
- b. a 0.0500 M sodium hydroxide solution.
- c. a 0.0500 M benzoic acid solution.
- d. a 0.0500 M sodium benzoate solution.

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in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

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Mastery Through Quizzing Stan Skrabut, 2025-03-24 Clear your desks You have a quiz Few phrases spark more anxiety in students The tension the stomach knots the panicked glances around the room We've all been there But what if quizzing wasn't something to dread What if it became the key to deeper learning greater confidence and real mastery instead Mastery Through Quizzing isn't just about testing knowledge but transforming how we learn Whether you're an educator administrator or instructional designer this book will show you how to turn low stakes assessments into powerful tools for engagement retention and long term success Discover how to Shift from high pressure exams to a growth focused quizzing strategy that reduces anxiety and boosts confidence Design questions that drive critical thinking not just memorization Leverage technology and test banks to make quizzes more effective and scalable Implement a step by step Mastery Quizzing Strategy to help students achieve real understanding Unlike traditional assessments that measure what students don't know mastery quizzing helps them build knowledge step by step reinforcing what they know and guiding them forward With decades of experience in instructional technology and education I've seen firsthand how this method can transform classrooms I'm sharing the approach to help you move beyond outdated testing models and create a learning experience that works It's time to rethink quizzing Let's turn it from a source of stress into a stepping stone for mastery Scroll up and grab your copy today

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Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS,2018-10-15 This guide is separated into first term and second term general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems

Broadening Participation in STEM Zayika Wilson-Kennedy,Goldie S. Byrd,Eugene Kennedy,Henry T. Frierson,2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines

Catalogue University of California, Santa Cruz, **ELECTRONIC STRUCTURE AND THE PERIODIC TABLE** STERLING. EDUCATION,2022 **Bulletin** ,1962 Science ,1969

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