

Reference Tables for Physical Setting/EARTH SCIENCE

Radioactive Decay Data

RADIOACTIVE ISOTOPE	DISINTEGRATION	HALF-LIFE (years)
Carbon-14	$^{14}\text{C} \rightarrow ^{14}\text{N}$	5.7×10^3
Potassium-40	$^{40}\text{K} \rightarrow ^{40}\text{Ar}$ $^{40}\text{K} \rightarrow ^{40}\text{Ca}$	1.3×10^9
Uranium-238	$^{238}\text{U} \rightarrow ^{206}\text{Pb}$	4.5×10^9
Rubidium-87	$^{87}\text{Rb} \rightarrow ^{87}\text{Sr}$	4.9×10^{10}

Specific Heats of Common Materials

MATERIAL	SPECIFIC HEAT (Joules/gram • °C)
Liquid water	4.18
Solid water (ice)	2.11
Water vapor	2.00
Dry air	1.01
Basalt	0.84
Granite	0.79
Iron	0.45
Copper	0.38
Lead	0.13

Equations

$$\text{Eccentricity} = \frac{\text{distance between foci}}{\text{length of major axis}}$$

$$\text{Gradient} = \frac{\text{change in field value}}{\text{distance}}$$

$$\text{Rate of change} = \frac{\text{change in value}}{\text{time}}$$

$$\text{Density} = \frac{\text{mass}}{\text{volume}}$$

Properties of Water

Heat energy gained during melting	334 J/g
Heat energy released during freezing	334 J/g
Heat energy gained during vaporization	2260 J/g
Heat energy released during condensation	2260 J/g
Density at 3.98°C	1.0 g/mL

Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

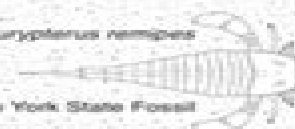
ELEMENT (symbol)	CRUST		HYDROSPHERE	TROPOSPHERE
	Percent by mass	Percent by volume	Percent by volume	Percent by volume
Oxygen (O)	46.10	94.04	33.0	21.0
Silicon (Si)	28.20	0.88		
Aluminum (Al)	8.23	0.48		
Iron (Fe)	5.63	0.49		
Calcium (Ca)	4.15	1.18		
Sodium (Na)	2.36	1.11		
Magnesium (Mg)	2.33	0.33		
Potassium (K)	2.09	1.42		
Nitrogen (N)				78.0
Hydrogen (H)			66.0	
Other	0.91	0.07	1.0	1.0

2011 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2011–12 school year. The first examination for which these tables will be used is the January 2012 Regents Examination in Physical Setting/Earth Science.

Eurypterus remipes

New York State Fossil



Practice Earth Science Reference Table Questions

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Table of Contents Practice Earth Science Reference Table Questions

1. Understanding the eBook Practice Earth Science Reference Table Questions
 - The Rise of Digital Reading Practice Earth Science Reference Table Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Practice Earth Science Reference Table Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Practice Earth Science Reference Table Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Practice Earth Science Reference Table Questions
 - Personalized Recommendations
 - Practice Earth Science Reference Table Questions User Reviews and Ratings
 - Practice Earth Science Reference Table Questions and Bestseller Lists

5. Accessing Practice Earth Science Reference Table Questions Free and Paid eBooks
 - Practice Earth Science Reference Table Questions Public Domain eBooks
 - Practice Earth Science Reference Table Questions eBook Subscription Services
 - Practice Earth Science Reference Table Questions Budget-Friendly Options
6. Navigating Practice Earth Science Reference Table Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Practice Earth Science Reference Table Questions Compatibility with Devices
 - Practice Earth Science Reference Table Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Practice Earth Science Reference Table Questions
 - Highlighting and Note-Taking Practice Earth Science Reference Table Questions
 - Interactive Elements Practice Earth Science Reference Table Questions
8. Staying Engaged with Practice Earth Science Reference Table Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Practice Earth Science Reference Table Questions
9. Balancing eBooks and Physical Books Practice Earth Science Reference Table Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Practice Earth Science Reference Table Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Practice Earth Science Reference Table Questions
 - Setting Reading Goals Practice Earth Science Reference Table Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Practice Earth Science Reference Table Questions
 - Fact-Checking eBook Content of Practice Earth Science Reference Table Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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