4

ATOMIC STRUCTURE

Vocabulary Review

Inhatels the correct receibulary term to each supplemed statement. While the latter of the curvest term on the late.

	Column A		Column B
	defined as /y the mass of a certice; 12 atoms	(36)	tectopes
	mentacl core of an atom, which constants most of the attent's mass	*	nectors
a.	a vertical column of elements in the periodic table	36	MAKE .
	substrenic particles with no charge	144	electrons.
	positively charged substantial particles	200	emploid despited
	the smallest particle of an element that setalos to identity in a characteristic reaction	*	scuriorio mass unit
	the number of protons to the endous of an element.	336	group.
	negotively charged reliescenic particles	366	madesa
	attens with the same number of protons but different combens of newtrons	4	periodic sable
10.	so arrangement of elements according to similarities	3.	protoco

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Zhe Xu

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