

CHEMISTRY SEMESTER 1 REVIEW QUESTIONS KEY

These questions are a small selection for each unit this semester. Refer to the exam scoring-guide and your self-reflection guide to determine which questions you should focus your studies on. Answers to select questions can be found at the end of this document.

Unit 1: Gas Laws

- A gas is heated in a closed, rigid vessel. What happens to the molecules inside?
The molecules start to move faster. They hit the inside of the container with greater force, which increases the pressure.
- A gas is compressed (volume decreased) at a constant temperature. What happens to the pressure inside the container? Explain your reasoning.
Since the volume is decreased, the molecules hit the sides of the container with increased frequency. This raises the pressure.
- When inflating a car tire, the pressure doubles from 200 to 400 kPa. Why would the pressure double when the volume is increasing?
The volume may be increasing, but it is because more molecules are being added to the system. More molecules means more collisions, which increases the pressure.
- What is the relation between temperature and pressure when volume is held constant?
With a constant volume, if temperature increases pressure increases as well. Think about heating an aerosol can (hair spray) – the volume is rigid and fixed. If you heat the can, the internal molecules move faster, hitting the sides harder and more frequently. This will increase the pressure so much that the can may explode.

Unit 2: Atomic structure and theory, periodic table

- How many protons, neutron, and electrons are in a neutral atom of Ag-108?
Protons = 47
Neutrons = 61
Electrons = 47

6. Fill in the table below.

Name	Sodium	Nitrogen
Symbol	$^{23}_{11}\text{Na}$	$^{14}_7\text{N}$
Atomic number	11	7
Mass number	23	15
Protons	11	7
Neutrons	12	8
Electrons in a neutral atom	11	7

- Analyze the following two isotopes. What is the same between both isotopes? What is different? $^{15}_7\text{N}$ and $^{14}_7\text{N}$
Both isotopes have the same number of protons (7) yet a different number of neutrons (8 and 7, respectively). The different number of neutrons yields a different mass number for the isotopes.
- Chlorine has 2 isotopes, Cl-35 and Cl-37. Cl-35 has an abundance of 75.77% while Cl-37 has an abundance of 24.23%. Will the atomic mass of chlorine be closer to 35 or 37? Explain your reasoning?
Atomic mass will be closer to Cl-35 because 'atomic mass' refers to the average mass of all isotopes of an element. If most chlorine isotopes have a mass of 35 while only a few have a mass of 37, the average mass will be closer to 35.
- Draw the Bohr model for sulfur. Label protons, neutrons, electrons, nucleus, and energy levels.



Prentice Hall Chemistry Chapter 1review Answers

**Thirumalaisamy P.
Velavancorresponding**



Prentice Hall Chemistry Chapter 1review Answers:

Prentice Hall Chemistry Chapter 1review Answers: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the engaging narratives that have captivated audiences this year. Prentice Hall Chemistry Chapter 1review Answers : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Prentice Hall Chemistry Chapter 1review Answers : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and gripping novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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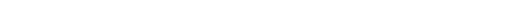
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