

Table 10.3

Comparison of Empirical and Molecular Formulas

Formula (name)	Classification of formula	Molar mass
CH	Empirical	13
C ₂ H ₂ (ethyne)	Molecular	26 (2 × 13)
C ₆ H ₆ (benzene)	Molecular	78 (6 × 13)
CH ₂ O (methanal)	Empirical and Molecular	30
C ₂ H ₄ O ₂ (ethanoic acid)	Molecular	60 (2 × 30)
C ₆ H ₁₂ O ₆ (glucose)	Molecular	180 (6 × 30)

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