

PHYSICAL AND CHEMICAL CHANGES LAB- virtual

Pre-Lab Discussion:

Chemistry is the study of matter and the changes it undergoes. These changes can be divided into 2 classes: physical changes and chemical changes.

In a physical change, one or more physical properties of a substance are altered. Examples of such physical properties are: size, shape, color, and state of matter. Grinding, melting, dissolving, and evaporating are all physical changes. No new substance or substances are formed as a result of a physical change.

A chemical change results in the formation of one or more "new" substances. These new substances differ in chemical properties and composition from the original substance. Iron rusting and paper burning are two examples of chemical change. 5 signs of a chemical change are formation of bubbles (gas produced), formation of a precipitate (solid formed), change of color, change of temperature or production of an odor.

This experiment will help you understand the differences between physical and chemical changes and to recognize each type of change when it occurs.

Station 1 video

1. Scrape off a small amount of candle wax into a glass container.
2. Heat the outside of the glass container with a flame lighter until the wax melts. [video for wax scrape and heat](#)
3. Light the candle and allow it to burn for 30 seconds. [video of candle burning](#)

	observations	P or C	Evidence to support P or C
candle	its white its a solid	P	because it's still wax but just smaller pieces
Scraping of candle wax	melted fast, liquid, clear	P	it's still wax but there was a small color change and changed to a liquid
Melting of candle wax	it burned quickly, where did the wax go?, as it burned it became smaller	C	not reversible, its a "new" substance
Burning of candle	same this ^	C	not reversible, its a "new" substance

Station 2

1. Tear a piece of paper into pieces.
2. Place the pieces of paper on wire gauze and ignite them with a match. [video for paper tear and burn](#)

	observations	P or C	Evidence to support P or C
Sheet of paper	was white		
Torn paper	white, smaller	P	small pieces but still paper
Burned paper	ashes, color change,	C	turned to ash, cant g back, smoke

Prentice Hall Chemistry Physical And Chemical Change

Raffaella Di Napoli



Prentice Hall Chemistry Physical And Chemical Change:

Whispering the Strategies of Language: An Psychological Quest through **Prentice Hall Chemistry Physical And Chemical Change**

In a digitally-driven earth where monitors reign great and quick interaction drowns out the subtleties of language, the profound strategies and emotional nuances concealed within words usually get unheard. Yet, nestled within the pages of **Prentice Hall Chemistry Physical And Chemical Change** a interesting fictional value pulsing with fresh emotions, lies an exceptional journey waiting to be undertaken. Penned by a talented wordsmith, this marvelous opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the fabric of each and every word. Within the emotional depths of the poignant evaluation, we will embark upon a genuine exploration of the book is core styles, dissect its charming writing design, and fail to the powerful resonance it evokes strong within the recesses of readers hearts.

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