

Name _____ Class _____ Date _____

Section 2–3 Carbon Compounds (pages 44–48)



Key Concept

- What are the functions of each group of organic compounds?

The Chemistry of Carbon (page 44)

- How many valence electrons does each carbon atom have? Each carbon atom has four electrons.
- What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms.

Macromolecules (page 45)

- Many of the molecules in living cells are so large that they are known as macromolecules.
- What is the process called by which macromolecules are formed? Polymerization
- When monomers join together, what do they form? Polymers
- What are four groups of organic compounds found in living things?
 - Carbohydrates
 - Lipids
 - Nucleic acids
 - Proteins

Carbohydrates (pages 45–46)

- What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbohydrates.
- Circle the letter of each sentence that is true about carbohydrates.
 - ☒ Starches and sugars are examples of carbohydrates.
 - ☒ Living things use them as their main source of energy.
 - ☐ The monomers in sugar polymers are starch molecules.
 - ☒ Plants and some animals use them for strength and rigidity.
- Single sugar molecules are also called monosaccharides.
- Circle the letter of each monosaccharide.

☒ galactose	☒ glucose
☐ glycogen	☒ fructose

Prentice Hall Chemistry Chapter 2 Worksheet Answers

CH Cherryholmes



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Table of Contents Prentice Hall Chemistry Chapter 2 Worksheet Answers

1. Understanding the eBook Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - The Rise of Digital Reading Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Personalized Recommendations
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers User Reviews and Ratings
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers and Bestseller Lists
5. Accessing Prentice Hall Chemistry Chapter 2 Worksheet Answers Free and Paid eBooks
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers Public Domain eBooks
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers eBook Subscription Services
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers Budget-Friendly Options

6. Navigating Prentice Hall Chemistry Chapter 2 Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers Compatibility with Devices
 - Prentice Hall Chemistry Chapter 2 Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Highlighting and Note-Taking Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Interactive Elements Prentice Hall Chemistry Chapter 2 Worksheet Answers
8. Staying Engaged with Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Chapter 2 Worksheet Answers
9. Balancing eBooks and Physical Books Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Chapter 2 Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Setting Reading Goals Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Fact-Checking eBook Content of Prentice Hall Chemistry Chapter 2 Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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